

Drug Metabolism and Disposition: the biological fate of chemicals

Editor
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In this Issue:

Human and Rat Liver *N*-Demethylase
Activities
Stereoselective Disposition of S-8666 in
Liver
Ether Cleavage of 3-Phenoxybenzoic Acid
Phase I and II Enzyme Expression in Skin
Purification of Bunitrolol 4-Hydroxylase from
Rats
Sulfate and PAPS Homeostasis
Product Inhibition of Debrisoquin Metabolism
Biotransformations of Verlukast
Kinetics of a CCK_B Receptor Antagonist
ST4-ddC Drug Interaction
Obesity and Gentamicin Toxicokinetics
Interleukin-1 β and P-450 in Fetal
Hepatocytes

Estrogen Sulfation
Tissue Distribution of CoQ₁₀
Aerosolized MK-679 in Rats
Pharmacokinetics of Amphotericin B
Dermal Absorption of 2- and 4-
Chloronitrobenzene in Rats
Enzymatic Hydrolysis of Allylic Epoxides
Modeling Cytochrome P-450 in the Brain
Isolation and Characterization of
Carbinolamides and Glucuronides
Influence of DB-cAMP in Rats
Carbonyl (Phenone) Reductase in Human
Liver
Distribution of TCDD in C57BL/6J Mice
New Metabolites of Bis(2-ethylhexyl)
Phthalate

INSTRUCTIONS TO AUTHORS

Submission of manuscripts. Drug Metabolism and Disposition will review *in vitro* and *in vivo* experimental results that contribute significant and original information on xenobiotic metabolism and disposition. The term xenobiotic includes pharmacologic agents as well as environmental chemicals. Pharmacokinetic and pharmacodynamic manuscripts and those involving mechanisms are invited. Manuscripts concerned with factors which affect the biological fate of chemicals such as genetic, nutritional or hormonal are of interest. Papers addressing toxicological consequences of xenobiotic metabolism are appropriate.

Three copies of each manuscript should be sent to Dr. Vincent G. Zannoni, Editor, *Drug Metabolism and Disposition*, Department of Pharmacology, University of Michigan Medical School, MSI, Ann Arbor, Michigan 48109-0626. FAX number: (313)-763-4450. Submission of a manuscript implies that the material contained therein has not previously been published except as an abstract for a scientific meeting, and that it is not being submitted elsewhere.

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A. Full-length papers should be arranged as follows:

1. *Title page*, containing the title of the paper, names of all authors, and the institution(s) where the work was done. The title should have no footnote numbers (see *Footnotes* below). The title should briefly yet explicitly indicate the contents of the paper. Names of chemicals or chemical classes studied, species used, etc., should be included in the title.

2. *Running title* not exceeding 50 total characters and spaces. The name and address of the person to whom editorial correspondence and galley proofs should be sent should appear at the bottom of this page.

3. *Abstract* of not more than 250 words.

4. *Introduction*. A brief summary of the pertinent literature and a statement of the aims of the work.

5. *Materials and Methods*. Species, strains, sexes, and ages or sizes of animals, with Latin names where required for distinction, should be given. Sources and purities of chemicals other than common reagents should be indicated. Equipment used and conditions of use should be specified. When published methods are used, a bibliographic reference is sufficient; minor modifications should be described. When a method has been extensively modified, the entire new procedure should be described. Authors should attempt to describe their work in all cases so that their peers would be able to repeat the experiments. Where conditions for similar experiments vary throughout the work, these may be indicated in legends to figures and tables. Properties and proof of structure must be given for reference compounds used for metabolite identification.

6. *Results*. These should be presented as much as possible in graphic and tabular form. When, however, a table would include only two or three values, it may be preferable to present the data in sentence form in the text. Authors should avoid using several tables describing very similar experiments; these should be combined wherever possible, unless this would result in overcomplicated, unwieldy tables. The same data should normally not be repeated in tables and figures. The text should be used to describe and summarize the data and to draw primary conclusions from them, but not to repeat the numerical data. No extended discussion of the results should be included in this section.

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Although it is normally preferable to separate the Results and Discussion sections, these sections, *e.g.*, when an extended discussion of some of the experiments is required for an understanding of subsequent experiments, may have to be combined occasionally.

8. *Acknowledgments* of technical assistance, gifts of materials, and other aid. Financial support should not be mentioned here, but rather in an unnumbered footnote to the title (see *Footnotes*, below).

9. *References*, numbered in order of citation in the text. Examples of style of references follow:

1. S. S. Lau, G. D. Abrams, and V. G. Zannoni: Metabolic activation and detoxification of bromobenzene leading to cytotoxicity. *J. Pharmacol. Exp. Ther.* **214**, 703–708 (1980).
2. T. C. Butler: The distribution of drugs. In “Fundamentals of Drug Metabolism and Drug Disposition” (B. N. La Du, H. G. Mandel, and E. L. Way, eds.), pp. 44–62. Williams & Wilkins, Baltimore, 1971.

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11. *Tables*, each on a separate page. Tables are to be numbered with arabic numbers. The title should be in italics (or underlined) with only the first word and proper names capitalized. General statements about the table should follow the title in paragraph form. Footnotes to the table should be indicated by italicized lower case superscript letters, starting with *a* for each table. Footnotes should be typed immediately below each table.

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13. *Index terms*. A list of index terms which may be used in constructing the annual index should constitute the last typed page of the manuscript.

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nm (not $m\mu$) wavelength]; ϵ (molar absorption coefficient, with units $\text{M}^{-1} \text{cm}^{-1}$); UV (ultraviolet); IR (infrared); ESR (electron-spin resonance); NMR (nuclear magnetic resonance); δ [chemical shift, with units ppm (parts per million)]; s (singlet); d (doublet); t (triplet); m (multiplet); amu (atomic mass units); m/z (mass/charge ratio).

Chromatography: TLC (thin-layer chromatography); R_F (retardation factor); GLC (gas-liquid chromatography); R_T (retention time); GC/MS (coupled gas chromatography-mass spectrometry); HPLC (high-pressure liquid chromatography).

Equilibrium and kinetic constants: K_d (dissociation constant); K_s or K_i (dissociation constant of enzyme-substrate or enzyme-inhibitor complex); K_M (Michaelis constant); V_{max} (maximum initial velocity); k (rate constant); $\text{p}K_a$ (negative logarithm of acidic dissociation constant); $t_{1/2}$, half-life; AUC, area under the curve of plasma concentrations vs. time.

Statistics: p (probability of chance observation); N (number of experiments); SD (standard deviation of the series); SE (standard error of the mean).

Other abbreviations: °C (degrees of temperature); g (acceleration due to gravity, as in 9000g); rpm (revolutions per min); LD_{50} and ED_{50} (median lethal and effective doses); iv (intravenous); ip (intraperitoneal); im (intramuscular); sc (subcutaneous); po (peroral); m.p. (melting point); sp.g. (specific gravity).

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DRUG METABOLISM AND DISPOSITION

The Biological Fate of Chemicals

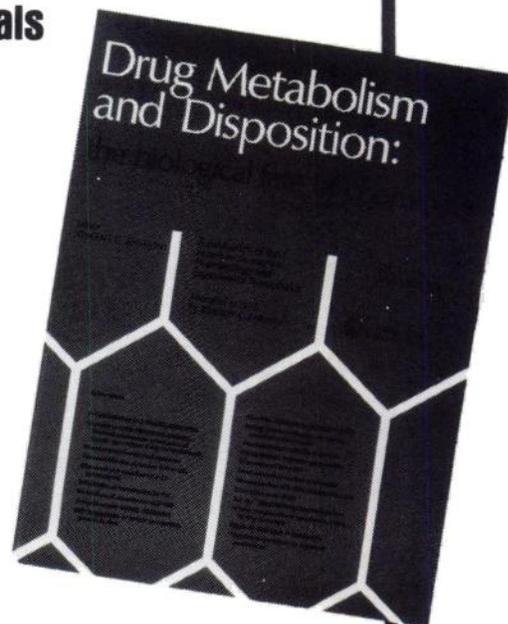
Editor: Vincent G. Zannoni, PhD, University of Michigan,
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DRUG METABOLISM AND DISPOSITION publishes *in vitro* and *in vivo* experimental results that bring readers significant and original information on xenobiotic metabolism and disposition, including metabolism of all pharmacologic agents or drugs and environmental chemicals, reactants, and preservatives. All papers are refereed to ensure a high standard of publication. The areas covered are:

- pharmacokinetics
- pharmacodynamics
- genetic, nutritional, and hormonal factors affecting the biological fate of chemicals
- toxicological consequences of xenobiotic metabolism

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