DISPOSITION AND METABOLISM OF OLANZAPINE IN MICE, DOGS, AND RHESUS MONKEYS

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(Received September 12, 1996; accepted January 30, 1997)

ABSTRACT:

Olanzpine (OLZ) is a novel antipsychotic agent with a high affinity for serotonin (5-HT2a) dopamine (D1/D2/D4), muscarinic (m1-m5), adrenergic (a1), and histamine (H1) receptors. The pharmacokinetics, excretion, and metabolism of OLZ were studied in CD-1 mice, beagle dogs, and rhesus monkeys after a single oral and/or intravenous dose of [14C]OLZ. After oral administration, OLZ was well absorbed in dogs (absolute bioavailability of 73%) and to the extent of at least 55% in monkeys and 32% in mice. The terminal elimination half-life of OLZ was relatively short in mice and monkeys (3–9 hr) and long in dogs (9 hr). In mice and dogs, radioactivity was predominantly eliminated in feces; but, in monkeys, the major route of elimination of radioactivity was urine. Dogs and monkeys excreted in urine, respectively, 38% and 55% of the dose over a 168-hr period, whereas the fraction of the dose excreted in urine of respectively, 38% and 55% of the dose over a 168-hr period, whereas the fraction of the dose excreted in urine of mice and dogs over the collection period (120 hr) was 32%. OLZ was subject to substantial first-pass metabolism; at the ratio of AUC OLZ to AUC radioactivity was, respectively, 10%, 14%, and 4% in mice, dogs, and monkeys. The principal urinary metabolites in mice were 7-hydroxy OLZ glucuronide, 2-hydroxymethyl OLZ, and 2-carboxy OLZ accounting for ~10%, 4%, and 2% of the dose. Metabolites that were present in urine in lesser amounts were 7-hydroxy OLZ, N-desmethyl OLZ, and N-desmethyl-2-hydroxymethyl OLZ. In dogs, the major metabolite accounting for ~8% of the dose was 7-hydroxy-N-oxide OLZ. Other metabolites identified were 2-hydroxymethyl OLZ, 2-carboxy OLZ, N-oxide OLZ, 7-hydroxy OLZ, and its glucuronide and N-desmethyl OLZ. The major metabolite in monkey urine was N-desmethyl-2-carboxy OLZ, and accounted for ~17% of the dose. In addition, N-oxide-2-hydroxymethyl OLZ, N-oxide-2-carboxy OLZ, N-desmethyl-2-hydroxymethyl, 2-carboxy OLZ, and 2-hydroxymethyl OLZ were identified in monkey urine. Thus, in mice and dogs, OLZ was metabolized through aromatic hydration, aliphatic oxidation, N-dealkylation, and N-oxidation reactions. In monkeys, OLZ was biotransformed mainly through double oxidation reactions involving the allylic carbon and methyl piperazine nitrogen. Whereas the oxidative metabolic profile of OLZ in animals was similar to that of humans, animals were notable for not forming appreciable amounts of the principal human metabolite (i.e. 10-N-glucuronide OLZ).

OLZ (fig. 1) is a new antipsychotic drug with a thienobenzodiazepinyl structure. OLZ displays a broad pharmacological profile with potent activity at dopamine (D1/D2/D4), serotonin (5-HT2a/2c), muscarinic (especially m1), histamine (H1) and adrenergic (a1) receptors (1, 2). The receptor binding profile of OLZ is very similar to clozapine, although OLZ is a more potent inhibitor of these receptors.

In clinical studies with patients suffering from schizophrenia or schizophreniform disorder, OLZ was effective in the treatment of both positive and negative symptoms of schizophrenia, with a low incidence of extrapyramidal side-effects (3–5). Antipsychotic efficacy of OLZ was demonstrated in the dose range of 5–20 mg/day.

The disposition and metabolism of OLZ after a single oral dose to healthy volunteers has recently been reported (6). OLZ was well absorbed and extensively metabolized. The primary metabolic route was N-glucuronidation. OLZ also underwent oxidative metabolism through N-oxidation, N-demethylation, and 2-alkyl hydroxylation. This study describes the comparative absorption, pharmacokinetics, and metabolism of OLZ in mice, dogs, and monkeys. The studies were conducted after the administration of [14C]OLZ.

Materials and Methods

Reference Compounds and Other Materials. The following compounds were synthesized at Lilly Research Laboratories: OLZ (2-methyl-4-(4-methyl-1-piperazinyl)-10H-thieno[2,3-B][1,5]benzodiazepine), [4,10a-14C]-OLZ ([14C]OLZ; radiochemical purity, 98.7%; specific activity, 26.2 μCi/mg), 4’-N-desmethyl OLZ (N-desmethyl OLZ, 2-methyl-4-(1-piperazinyl)-10H-thieno[2,3-B][1,5]benzodiazepine), 4’-N-oxide OLZ (N-oxide OLZ, 4’-methyl-10H-thieno[2,3-B][1,5]benzodiazepine, 4’-methyl-10H-thieno[2,3-B][1,5]benzodiazepine-4-yl)-1-methylpiperazine-1-oxide), 2-hydroxy-10H-OLZ (4-(4-methyl-1-piperazinyl)-10H-thieno[2,3-B][1,5]benzodiazepine-2-methanol), 2-carboxymethyl OLZ (methyl 4-(4-methyl-1-piperazinyl)-10H-thieno[2,3-B][1,5]benzodiazepine-2-carboxylate), 7-ethoxy OLZ (7-ethoxy-2-methyl-4-(4-methyl-1-piperazinyl)-10H-thieno[2,3-B][1,5]benzodiazepine), and 4’-N-desmethyl-2-hydroxymethyl OLZ (4-(1-piperazinyl)-10H-thieno[2,3-B][1,5]benzodiazepine-2-methanol). NEE-154 Glusulase was purchased from the DuPont Company (Wilmington, DE). β-Saccharolactone was purchased from Sigma Chemical Co. (St. Louis, MO). Scintisol was supplied by Isolab, Inc. (Akron, OH). HPLC-grade ammonium acetate, acetonitrile, triethylamine, and reagent-grade boron tribromide were purchased from Fisher Scientific (Fair Lawn, NJ). 7-Hydroxy OLZ was pre-
pared by deethylation of 7-ethoxy OLZ, and 2-carboxyl OLZ was prepared by hydrolysis of 2-carboxymethyl OLZ. Approximately 2 mg of each starting material was placed in separate siliconized tubes and dissolved in methylene chloride (2 ml). The solution was flushed with nitrogen and treated with boron tribromide solution (2 ml of 25% solution in methylene chloride). The reaction was allowed to proceed at room temperature for 2 hr. Approximately 90% of 7-ethoxy OLZ was converted to 7-hydroxy OLZ, whereas ~50% of 2-carboxymethyl was converted to the corresponding acid as determined by HPLC and electrospray LC/MS. N-Desmethyl-2-carboxy OLZ was prepared by oxidizing the corresponding hydroxy compound using chromium trioxide (7).

Animal Experiments. All animal experiments were conducted according to protocols approved by the Eli Lilly Animal Care and Use Committee. The dosing solution used for all animal studies was prepared by dissolving the required amounts of OLZ and [14C]OLZ in 1 M HCl and titrating the solution to a pH of 6 by the addition of 0.1 M NaOH. The appropriate volume was then obtained by the subsequent addition of water.

Mouse. Male CD-1 mice were obtained from Charles River Laboratories (Wilmington, MA) and acclimatized for 3 days before use. Food and water were supplied ad libitum at all times throughout the experiment. For radiolabeled excretion study, the mice were divided into three groups, with each group containing five mice. Each animal was administered a single oral (gavage) dose of OLZ (15 mg/kg containing 420 mCi/kg of [14C]OLZ). Blood was collected and pooled from four mice at 0.5, 1, 2, 4, 7, 12, 24, 48, and 72 hr after the dose. Plasma was obtained by centrifugation and stored at -70°C until analysis. Metabolite identification was conducted in urine obtained from mice given a 20 mg/kg dose. Urine samples collected for 24 hr postdose from eight mice were combined and stored at -70°C until analyzed.

Dog. Four female beagle dogs (age: 2–4 years; weight: 8.7–13.1 kg) were obtained from stock animals maintained at Lilly Research Laboratories and placed in individual stainless-steel metabolism cages. Animals were fasted overnight before and 2 hr after drug administration. Animals were given a single oral (gavage) dose of OLZ (5 mg/kg containing 420 mCi/kg of [14C]OLZ). Urine and fecal samples were collected at 24-hr intervals for up to 120 hr. For pharmacokinetic study, four mice were used for each time point and dosed as described. Blood was collected and pooled from four mice at 0.5, 1, 2, 4, 7, 12, 24, 48, and 72 hr after the dose. Plasma was obtained by centrifugation and stored at -70°C until analysis. Metabolite identification was conducted in urine obtained from mice given a 20 mg/kg dose. Urine samples collected for 24-hr postdose from eight mice were combined and stored at -70°C until analyzed. Additional mice (10) were administered a single oral dose (15 mg/kg) of OLZ, and plasma was collected at ~1 hr for metabolite identification.

Hydrolysis of Conjugates. Glucuronide conjugates (~2 μg) isolated from urine were hydrolyzed to the corresponding aglycone by incubating with Glusulase (containing 2,070 units of β-glucuronidase and 150 units sulfatase) at 37°C for up to 20 hr. Incubations were also conducted in the absence of Glusulase and in the presence of β-saccharolactone (0.0325 M).

HPLC Separation of Metabolites. The HPLC system consisted of a Beckman pump, NEC controller, Waters Wisp autosampler, Applied Biosystems UV detector, and a Berthold radiodetector with 150 μl yttrium solid cell. Aliquots (~200 μl) of concentrated urine or extract were analyzed on a Hypersil C18 column (5 μm particle size, 0.46 × 25 cm) using a gradient containing A (0.1 M ammonium acetate) and B (1% triethylamine in acetonitrile). The initial solvent composition was 90% A and 10% B. After 2 min, the pump was programmed to increase solvent B by 2.5%/min until a proportion of 40% A and 60% B was achieved. The mobile phase was maintained for 8 min at that composition. The flow rate was 1 ml/min. Metabolites were isolated by mixed-mode, solid-phase extraction, separated with a reversed-phase method, and detected electrochemically. The upper and lower limits of quantitation of the assay were 100 and 1 ng/ml, respectively.
by collecting the radioactive eluent as it eluted off the column. Several
injections were made to obtain a sufficient amount of each metabolite for mass
spectral identification.

**LC-MS/MS.** Isolated metabolites were analyzed by LC/MS and LC-
MS/MS on a Finnigan MAT TSQ700. Metabolites were introduced into the
electrospray LC interface using a Waters Model 600 pump. Metabolites were
separated on an Inertsil C18 column (5 μm particle size, 0.46 x 25 cm) using
the same gradient as described with 0.05 M ammonium acetate and acetonitrile.
Injection volumes ranged from 10 to 200 μl. The flow rate was 1 ml/min,
and the effluent was split such that equal volumes were delivered into the ion
source and a Raytest Ramona model 5LS radiodetector. MS spectra were
obtained by scanning from m/z 20 to 600 every second. For CID experiments,
the collision gas (argon) pressure was maintained at 2.0 m torr, and the
collision offset voltage was −20 eV. MS and MS/MS spectra were averaged
for 1 min.

**NMR Spectroscopy.** Proton and carbon-13 NMR spectra were recorded in
d6-DMSO or CDC13 on a Bruker AMX spectrometer operating at 500 MHz.
Chemical shifts are reported in ppm relative to tetramethylsilane.

**Results**

**Excretion of Radioactivity.** Mice administered a single oral dose
(15 mg/kg) of OLZ eliminated 64.3 ± 3.4% (mean ± SD) and 31.9 ±
2.8% of the radioactivity, respectively, in feces and urine over a
120-hr period (table 1). The majority of the dose (>78%) was excreted
during the first 48 hr of dosing. Less than 1% of the administered
dose was recovered in the carcasses.

In dogs, ~84% of the radioactivity was recovered after 168 hr, with
slightly more radioactivity eliminated in the feces (45.6 ± 5.4%) than
in the urine (38.4 ± 2.6%). Greater than 50% of the dose was
recovered within 48 of dosing (table 1).

In monkeys, renal excretion was the primary mode of radiocarbon
elimination accounting for 54.6 ± 3.7% of the dose. Another 28.5 ±
5.2% of the dose was eliminated via the feces over the same period.
Greater than 50% of the dose was eliminated in the urine and feces 24
hr after the dose (table 1). There was no difference between males
and females with respect to the amount of radioactivity in either the urine
or feces.

**Pharmacokinetics.** Mice. Pharmacokinetic parameters of OLZ and
radioactivity in mice are shown in table 2. OLZ was quantitated in
plasma using an HPLC assay with a lower limit of quantitation of 1
ng/ml. The Cmax of OLZ was 421 ng/ml and occurred at 0.5 hr after
dosage. The corresponding value for radioactivity was 2,260 ng-
eq/ml and was reached at a much later time (4 hr). At 0.5 hr, OLZ
accounted for ~19% of plasma radioactivity. This is indicative of the
two-phase (fig. 3), with the terminal phase displaying a half-life of
12.0 hr. The ratio of AUC OLZ to AUC for OLZ were, respectively, 871
and 757 ng-eq * hr/ml. Thus, the mean elimination half-life of OLZ was 3.7 hr. The elimination of OLZ from plasma seemed to be
more radioactivity, compared with a value of 18% after an oral dose.

The mean Cmax of OLZ was 172 ± 69 ng/ml and occurred between
1 and 3 hr in 3 of the 4 animals tested. The fourth animal had a
Cmax of 6 hr. The elimination of OLZ from plasma seemed to be
more radioactivity, compared with a value of 18% after an oral dose.

The mean Cmax for radioactivity in plasma was 1 ± 0.0 hr, and the
Cmax for radioactivity was 949 ± 296 ng-eq/ml. Plasma radioactivity declined with a
mean half-life of 27.6 ± 12.0 hr. The ratio of AUC OLZ to AUC
radioactivity was 0.14.

After a single IV dose of OLZ to three dogs, the mean Cmax and
AUC for OLZ were, respectively, 871 ± 241 ng/ml and 2,633 ±
1,041 ng * hr/ml. The corresponding values for plasma radioactivity
were 1,145 ± 195 ng-eq/ml and 18,813 ± 2,598 ng-eq * hr/ml. Thus,
after an IV administration, at the tmax OLZ accounted for 76% of the
radioactivity, compared with a value of 18% after an oral dose.

Because the amount of radioactivity excreted in urine after the oral
and IV doses was almost the same (38.4% and 39.7% of the dose), the
decreased bioavailability after oral administration is likely due to the
first-pass metabolism. The ratio of AUC OLZ to 14C AUC was the
same as that obtained after oral dosing. The absolute oral bioavail-
ability of OLZ was calculated to be 73%.

**Monkeys.** The mean Cmax of OLZ and radioactivity were 60 ± 18
and 757 ± 169 ng eq/ml, and were reached on average within 1.5 hr
postdose. Therefore, at the Cmax OLZ accounted for ~8% of the
plasma radioactivity. On the basis of AUC, the fraction of plasma
radioactivity represented by OLZ was ~4%.

The mean elimination half-life of OLZ was 3.4 ± 1.2 hr. The

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**TABLE 1**

**Urinary and fecal elimination of radioactivity in mice, dogs, and monkeys after the administration of a single oral dose of 14C-OLZ**

<table>
<thead>
<tr>
<th>Time (hr)</th>
<th>Mouse</th>
<th>Dog</th>
<th>Monkey</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Urine</td>
<td>Feces</td>
<td>Urine</td>
</tr>
<tr>
<td>0–24</td>
<td>25.0 ± 1.8</td>
<td>46.7 ± 2.9</td>
<td>20.5 ± 3.7</td>
</tr>
<tr>
<td>24–48</td>
<td>4.0 ± 1.1</td>
<td>11.6 ± 3.0</td>
<td>11.7 ± 5.1</td>
</tr>
<tr>
<td>48–120</td>
<td>2.9 ± 2.2</td>
<td>6.0 ± 2.3</td>
<td>5.6 ± 1.0</td>
</tr>
<tr>
<td>120–168</td>
<td>NC</td>
<td>NC</td>
<td>0.60 ± 0.13</td>
</tr>
<tr>
<td>Total</td>
<td>31.9 ± 2.8</td>
<td>64.3 ± 3.4</td>
<td>38.4 ± 2.6</td>
</tr>
<tr>
<td>Grand total</td>
<td>97.4 ± 1.1</td>
<td>84.3 ± 5.0</td>
<td>83.1 ± 3.9</td>
</tr>
</tbody>
</table>

Data represent mean ± SD.

a N = 3.

b % dose recovered from the carcass was 0.6%.

c N = 4.

d N = 4.

f NC, not collected.

f Includes radioactivity from cage washings and carcass.

s Includes cage washings.
elimination of radioactivity from plasma was biphasic (fig. 4), with the initial and terminal phases having half-lives of, respectively, 5.3 ± 0.7 and 98.7 ± 26.5 hr.

In Vitro Plasma Protein Binding. The plasma protein binding of OLZ was similar in the three species studied, with mean binding being 77%, 75%, and 83% in mice, dogs, and monkeys, respectively. The binding was concentration-independent (10 –1,000 ng/ml). The extent of protein binding was lower in these species than that reported for humans at 93% (6).

Metabolism. Mice. Upon partitioning pooled urine (0–24 hr) between ethyl acetate and water, 19% of the radioactivity was extracted into the ethyl acetate, whereas 76% remained in the aqueous fraction. An aliquot of the aqueous fraction was separated by HPLC with radiochemical detection and yielded the chromatogram in fig. 5. The corresponding HPLC chromatogram from the ethyl acetate extract is shown in fig. 6. The individual peaks were collected and analyzed by direct infusion electrospray MS and MS/MS. The following metabolites were identified in urine of mice by comparing their LC and LC-MS/MS properties to those obtained from synthetic standards. The metabolite that eluted as peak 1 in fig. 5 was identified as 2-carboxy OLZ on the basis of the similarity of its HPLC retention time and product ion spectrum to that obtained from a sample of synthetic 2-carboxy OLZ. The positive ion electrospray mass spectrum of the major urinary metabolite (peak 2, fig. 5) exhibited an MH+ ion at m/z 505, which suggested that the metabolite was the glucuronide of a hydroxylated OLZ derivative (M, OLZ = 312). The product ion spectrum of m/z 505 was dominated by the fragment at m/z 329, which is likely due to loss of dehydroglucuronic acid from the conjugate. β-Glucuronidase hydrolysis of the conjugate resulted in 7-hydroxy OLZ, confirming the major metabolite in urine as 7-hydroxy OLZ glucuronide. Peaks 3 and 4 were characterized as N-desmethyl-2-hydroxymethyl OLZ and 2-hydroxymethyl OLZ, respectively, by comparison with authentic standards. Six metabolites (fig. 6) were isolated from the ethyl acetate extract for MS identification. 2-Hydroxymethyl OLZ, which was also present in the aqueous fraction, was identified as the component eluting as peak 1 in fig. 6. The metabolite that eluted as peak 4 had the same HPLC retention volume and MS/MS fragmentation as authentic 7-hydroxy OLZ. The metabolite shown as peak 3 (fig. 6) was identified as N-desmethyl OLZ. Unchanged OLZ was also excreted in urine (peak 4, fig. 6). The identities of the other radiolabeled components in fig. 6 were not assigned.

TABLE 2

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Mouse&lt;sup&gt;a&lt;/sup&gt;</th>
<th>Dog</th>
<th>Monkey</th>
</tr>
</thead>
<tbody>
<tr>
<td>Dose (mg/kg)</td>
<td>15</td>
<td>5</td>
<td>5</td>
</tr>
<tr>
<td>C&lt;sub&gt;max&lt;/sub&gt; (ng or ng-eq/ml)</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>OLZ</td>
<td>421</td>
<td>172 ± 69</td>
<td>60 ± 18</td>
</tr>
<tr>
<td>14C</td>
<td>2,260</td>
<td>949 ± 296</td>
<td>757 ± 169</td>
</tr>
<tr>
<td>OLZ as % of 14C</td>
<td>19%</td>
<td>18%</td>
<td>8%</td>
</tr>
<tr>
<td>t&lt;sub&gt;max&lt;/sub&gt; (hr)</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>OLZ</td>
<td>0.5</td>
<td>3.3 ± 2.1</td>
<td>1 ± 0.0</td>
</tr>
<tr>
<td>14C</td>
<td>4</td>
<td>1 ± 0.0</td>
<td>1.5 ± 1.7</td>
</tr>
<tr>
<td>t&lt;sub&gt;1/2&lt;/sub&gt; (hr)</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>OLZ</td>
<td>3.2 (7–12 hr)&lt;sup&gt;b&lt;/sup&gt;</td>
<td>9.2 ± 1.4 (3–48 hr)&lt;sup&gt;b&lt;/sup&gt;</td>
<td>3.4 ± 1.2 (1–12 hr)</td>
</tr>
<tr>
<td>14C</td>
<td>10.6 (7–48 hr)&lt;sup&gt;b&lt;/sup&gt;</td>
<td>27.6 ± 12.0 (24–96 hr)&lt;sup&gt;b&lt;/sup&gt;</td>
<td>5.3 ± 0.7 (4–12 hr)</td>
</tr>
<tr>
<td>AUC (ng or ng-eq · hr/ml)</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>OLZ</td>
<td>1,522</td>
<td>1,923 ± 325</td>
<td>536.6 ± 208.3</td>
</tr>
<tr>
<td>14C</td>
<td>15,201</td>
<td>13,405 ± 2,123</td>
<td>14,429 ± 1,572</td>
</tr>
<tr>
<td>OLZ as % of total 14C</td>
<td>10%</td>
<td>14%</td>
<td>4%</td>
</tr>
</tbody>
</table>

<sup>a</sup> Mouse data obtained from pooled plasma, for dogs and monkeys values represent mean ± SD (N = 4).

<sup>b</sup> Points used in the determination of the respective half-lives.

**FIG. 2.** Plasma concentration vs. time profiles for OLZ and total radioactivity in mice given an oral dose of 15 mg/kg of [14C]OLZ. Radioactivity is expressed as mean ± SD (N = 4), whereas OLZ concentrations were obtained from pooled plasma.

**FIG. 3.** Plasma concentration vs. time profiles for OLZ and total radioactivity in dogs given an oral dose of 5 mg/kg of [14C]OLZ. Data are expressed as mean ± SD (N = 4).
confirmed, although MS/MS fragmentation indicated that they were OLZ metabolites. The peak eluting between peaks 2 and 3 had an apparent MH$^+$ ion of m/z 327, 2 Da less than that of 2-hydroxymethyl OLZ. This metabolite could possibly be the precursor of 2-carboxy OLZ, the 2-formyl derivative of OLZ.

In plasma, in addition to the parent compound, 2-hydroxymethyl OLZ, N-desmethyl OLZ and the glucuronide of a hydroxy OLZ metabolite were detected. Also, mass spectral data was obtained that indicated the presence of two isomeric glutathione conjugates of OLZ. The conjugates exhibited an MH$^+$ ion at m/z 618, which upon CID fragmentation, gave MH-129$^+$—a characteristic loss of glutathione conjugates (9), in addition to other fragments consistent with the glutathione conjugate of OLZ. The MS/MS data also suggested that the glutathione moiety was attached to one of the carbons of the benzene ring of OLZ.

Six metabolites of OLZ were identified in the urine of mice in addition to the parent compound. Based on the percentage of the radioactivity that was extracted into the ethyl acetate (19%) and the percentage remaining in the aqueous fraction (76%), the amount of urinary radioactivity accounted for by each metabolite was estimated as shown in table 3.

Dogs. Pooled urine sample from the first 48 hr after dosing was used for metabolite identification. The partitioning of radioactivity between ethyl acetate and water was similar to that obtained for mouse urine with 15% extracted into ethyl acetate and 79% of the radioactivity remaining in the aqueous fraction. The HPLC separation of the radioactive components in the ethyl acetate and aqueous fractions is shown, respectively, in figs. 7 and 8. In the ethyl acetate extract, 2-hydroxymethyl OLZ, N-oxide OLZ, 7-hydroxy OLZ, N-desmethyl OLZ (peaks 1–4; fig. 7) were identified in addition to the parent compound.

The major component in the aqueous fraction (peak 3, fig. 8) had a retention time that was different from the available standards. The electrospray MS of this metabolite gave an apparent protonated molecular ion of m/z 345. MS/MS experiments indicated that the metabolite was an N-oxygenated species with a hydroxyl group on the benzodiazepine moiety. Approximately 30 mg of the metabolite was isolated from urine using XAD-2 chromatography and further purified by HPLC fractionation. The $^1$H- and $^{13}$C-NMR data obtained for the metabolite are shown in table 4. The $^1$H-NMR of the metabolite was isolated from urine using XAD-2 chromatography and further purified by HPLC fractionation. The $^1$H- and $^{13}$C-NMR data obtained for the metabolite show in table 4. The $^1$H-NMR of the metabolite showed a downfield shift of the 4'-CH$_3$ to $\delta$ 3.08 ($\delta$ 2.21 for OLZ) and was identical to the value obtained for N-oxide OLZ. Similarly, the $^{13}$C-NMR exhibited a downfield shift of the 4'-CH$_3$ resonance to $\delta$ 58.09 ($\delta$ 45.73 for OLZ). Two-dimensional nuclear Overhauser enhancement was used to confirm the exact position of the hydroxyl group on the benzene ring of OLZ. The absence of a C-7 proton ($\delta$ 6.83 for OLZ) and the fact that the C-9 proton showed ortho coupling only ($J = 8$ Hz) indicated the hydroxyl group was at the C-7 position. Thus, on the basis of combined MS and NMR data, the major urinary metabolite in dogs was identified as 7-hydroxy-N-oxide OLZ.
The aqueous fraction also contained a metabolite (peak 1, fig. 8) that was identified as 2-carboxy OLZ. In addition to 7-hydroxy OLZ glucuronide (peak 2, fig. 8), a glucuronide of OLZ was identified (peak 4, fig. 8) and characterized as the tertiary N-glucuronide, OLZ 10-N-glucuronide.

LC/MS analysis of the XAD-2 extract indicated the presence of apparent protonated molecular ions at \(m/z\) 432 and 448. The product ion spectra of these metabolites indicated that the metabolites might be the cysteine adducts of OLZ and N-oxide OLZ. CID analysis of the ion at \(m/z\) 432 resulted in fragment ions at \(m/z\) 345 and 311 that could be produced, respectively, from loss of 87 Da as a neutral [CH\(_2\)-C-(NH\(_2\))COOH] from the cysteinyl moiety of the conjugate and complete cleavage of the cysteine residue. The ion at \(m/z\) 311 further fragmented to an ion at \(m/z\) 254. This transition is characteristic of OLZ and metabolites (6) and results from loss of 57 Da as CH\(_2\)-CH—NH—CH\(_3\) from the methyl piperazine ring of the molecules. MS/MS analysis (precursor \(m/z\) 448) of the putative cysteine conjugate of N-oxide OLZ yielded a fragmentation pattern that was different from that obtained for the corresponding conjugate of OLZ. The fragment at \(m/z\) 261 perhaps resulted from the combined loss of 100 Da (scission of the methyl piperazine ring) and 87 Da [CH\(_2\)-C-(NH\(_2\))COOH, from the cysteine residue]. The additional loss of possibly hydrogen sulfide resulted in a fragment at \(m/z\) 228. A weak ion at \(m/z\) 401 resulted from the loss of 47 Da from the methyl piperazine portion of the molecule that is a characteristic fragmentation pathway of N-oxide OLZ (6).

After 3 hr postdose, plasma contained OLZ, 2-hydroxymethyl, N-oxide, N-desmethyl, and the 7-hydroxy metabolites, as well as the glucuronide of 7-hydroxy OLZ. After 12 hr, the plasma metabolite profile was similar to that obtained at 3 hr, except that the level of N-oxide was lower than that of the 7-hydroxy metabolite and no 7-hydroxy glucuronide was detected.

The relative amount of each metabolite and parent drug in urine was estimated by HPLC with radiochemical detection as detailed in the Materials and Methods and is presented in table 3. The amount of 7-hydroxy OLZ was estimated from the ethyl acetate extract. The ethyl acetate extract contained 7-hydroxy OLZ; however, this metabolite was not detectable in the XAD-2 extract. The 7-hydroxy metabolite is fairly susceptible to air oxidation and could have decomposed during the lengthy XAD-2 extraction procedure.

**Monkeys.** An aliquot of the first 24-hr urine sample from each

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**TABLE 3**

<table>
<thead>
<tr>
<th>Compound</th>
<th>% Urinary Radioactivity</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Mouse</td>
</tr>
<tr>
<td>7-Hydroxy glucuronide</td>
<td>39.6</td>
</tr>
<tr>
<td>7-Hydroxy</td>
<td>1.1</td>
</tr>
<tr>
<td>7-Hydroxy-N-oxide</td>
<td>ND</td>
</tr>
<tr>
<td>N-desmethyl-2-hydroxymethyl</td>
<td>1.7</td>
</tr>
<tr>
<td>N-desmethyl-2-desmethyl-2-carboxy</td>
<td>ND</td>
</tr>
<tr>
<td>2-Hydroxymethyl</td>
<td>14.9</td>
</tr>
<tr>
<td>2-Hydroxymethyl glucuronide</td>
<td>ND</td>
</tr>
<tr>
<td>2-Carboxy</td>
<td>6.6</td>
</tr>
<tr>
<td>N-oxide</td>
<td>ND</td>
</tr>
<tr>
<td>N-oxide-2-hydroxymethyl</td>
<td>ND</td>
</tr>
<tr>
<td>N-oxide-2-carboxy</td>
<td>ND</td>
</tr>
<tr>
<td>N-desmethyl</td>
<td>1.6</td>
</tr>
<tr>
<td>CYS' conjugate</td>
<td>ND</td>
</tr>
<tr>
<td>CYS conjugate of N-oxide</td>
<td>ND</td>
</tr>
<tr>
<td>NAC conjugate</td>
<td>ND</td>
</tr>
<tr>
<td>Olanzapine</td>
<td>3.6</td>
</tr>
<tr>
<td>Total</td>
<td>69.1</td>
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</table>

*7-Hydroxy glucuronide coeluted with the cysteine conjugate of OLZ under the HPLC conditions used. An estimate for these metabolites was obtained using ms and comparing the ion intensities at \(m/z\) 432 and 505.

+ ND, not detected.
+ 2-Hydroxymethyl glucuronide coeluted with 2-carboxy. The level of each metabolite was estimated by ms on the basis of ion intensities at \(m/z\) 343 and 505.
+ 2-Carboxy coeluted with the cysteine conjugate of N-oxide. The relative amount of each metabolite was estimated by ms on the basis of ion intensities at \(m/z\) 343 and 448.
+ CYS, cysteine.
+ NAC, N-acetylcysteine.
Fig. 7. HPLC radiochromatogram obtained from the ethyl acetate extract of dog urine.
1, 2-hydroxymethyl OLZ; 2, N-oxide OLZ; 3, 7-hydroxy OLZ; 4, N-desmethyl OLZ; 5, OLZ.

Fig. 8. HPLC radiochromatogram of the aqueous fraction of dog urine.
1, 2-carboxy OLZ and the cysteine conjugate of N-oxide OLZ; 2, 7-hydroxy OLZ glucuronide and the cysteine conjugate of OLZ; 3, 7-hydroxy-N-oxide OLZ; 4, OLZ 10-N-glucuronide; 5, N-oxide OLZ.
monkey was concentrated ~10-fold and used for metabolite identification. A typical HPLC radiochromatogram obtained from the concentrated urine from one animal is shown in fig. 9. The following metabolites were identified in monkey urine by LC-MS/MS and in comparison with authentic standards: N-desmethyl-2-carboxy OLZ (peak 2, fig. 9), N-oxide-2-hydroxymethyl OLZ (peak 4, fig. 9), N-desmethyl-2-hydroxymethyl OLZ (peak 6, fig. 9), and 2-hydroxymethyl OLZ (peak 7, fig. 9).

LC/MS analysis of the metabolite eluting as peak 3 (fig. 9) produced an apparent protonated molecular ion of \( m/z \) 505, the CID of which produced major ions at \( m/z \) 329, 272, and 84. Loss of 176 Da as a neutral to produce the fragment at \( m/z \) 329 suggested that the metabolite was the glucuronide of a hydroxylated OLZ derivative. The isolated material was hydrolyzed to 2-hydroxymethyl OLZ in the presence of Glusulase, confirming this metabolite as the glucuronide conjugate of 2-hydroxymethyl OLZ. The enzyme hydrolysis to the aglycone was inhibited by \( \beta \)-sacchrolactone. 2-Carboxy OLZ coeluted with the glucuronide of 2-hydroxymethyl OLZ (peak 3, fig. 9) under the HPLC conditions used. The 2-carboxy metabolite was identified after \( \beta \)-glucuronidase treatment of a sample of the material that eluted as peak 3 (fig. 9), which resulted in the production of 2-hydroxymethyl OLZ, a compound with a substantial difference in retention time from that of 2-carboxy OLZ.

The HPLC radiochromatogram from urine contained a metabolite (peak 1, fig. 9) with a retention time that was different from any of the available synthetic metabolite standards. LC/MS analysis of the metabolite yielded an apparent protonated molecular ion of \( m/z \) 359 that afforded a product ion spectrum with major ions at \( m/z \) 312, 272, 259, 243, 199, 153, 135, and 85. The characteristic loss of 47 Da to produce the fragment at \( m/z \) 312 indicated this metabolite was an analog of \( N \)-oxide OLZ (6). The additional 30 Da (\( M_r \) of \( N \)-oxide OLZ = 328) suggested that the methyl group on the thiophene ring was oxidized to the carboxyl derivative. The fragment at \( m/z \) 199 is indicative of the presence of a carboxyl moiety at this position (unpublished observation). Thus, based on its distinct product ion spectrum and HPLC order of elution, this metabolite was tentatively assigned the structure of \( N \)-oxide-2-carboxy OLZ.

The HPLC chromatogram also contained another metabolite (peak 5, fig. 9) that did not match the LC-MS/MS properties of any available standards. Analysis of the metabolite by LC/MS afforded an apparent protonated molecular ion at \( m/z \) 474, the CID of which yielded a spectrum with major ions at \( m/z \) 345, 311, 288, 254, and 84. This product ion spectrum suggested that the metabolite was an \( N \)-acetylcysteine conjugate of OLZ. The ions at \( m/z \) 345 and 311 are

<table>
<thead>
<tr>
<th>Site</th>
<th>Chemical Shift (( \delta ))</th>
<th>( ^1H )</th>
<th>( ^13C )</th>
</tr>
</thead>
<tbody>
<tr>
<td>2—C=O</td>
<td>—</td>
<td>127.89</td>
<td></td>
</tr>
<tr>
<td>2—CH3</td>
<td>2.27</td>
<td>15.11</td>
<td></td>
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<tr>
<td>3—CH3</td>
<td>6.38</td>
<td>122.34</td>
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<tr>
<td>3(a)—C=O</td>
<td>—</td>
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</tr>
<tr>
<td>4—C=O</td>
<td>—</td>
<td>156.80</td>
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<tr>
<td>5(a)—C=O</td>
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<td>137.16</td>
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</tr>
<tr>
<td>6—CH3</td>
<td>6.27</td>
<td>113.83</td>
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<tr>
<td>7—C=O</td>
<td>—</td>
<td>155.15</td>
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</tr>
<tr>
<td>8—CH3</td>
<td>6.28</td>
<td>110.50</td>
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<td>9—CH3</td>
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<td>119.23</td>
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<tr>
<td>9(a)—C=O</td>
<td>—</td>
<td>141.01</td>
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</tr>
<tr>
<td>10—NH—</td>
<td>7.40</td>
<td>—</td>
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</tr>
<tr>
<td>10(a)—C=O</td>
<td>—</td>
<td>153.80</td>
<td></td>
</tr>
<tr>
<td>2′1′—CH2—</td>
<td>3.82; 3.52</td>
<td>41.02</td>
<td></td>
</tr>
<tr>
<td>3′5′—CH2—</td>
<td>a</td>
<td>63.78</td>
<td></td>
</tr>
<tr>
<td>4—CH3</td>
<td>3.08</td>
<td>58.09</td>
<td></td>
</tr>
</tbody>
</table>

\( a \) Difficult to observe because of the \( H_2O \) peak.
likely formed as a result of loss of 129 Da [CH₂A—(NHCOCH₃)—COOH] from the cysteinyl moiety of the metabolite and complete cleavage of the cysteinyl residue, respectively. The ion at m/z 288 seems to result from combined loss of 57 Da [CH₂ACH—NH—CH₃] from the methyl piperazine ring and [CH₂A—(NHCOCH₃)—COOH] from the cysteinyl moiety. The ion at m/z 311 further fragmented by losing 57 Da to give the ion at m/z 254; a typical fragmentation pathway for OLZ and metabolites (6).

The level of each metabolite was estimated by HPLC, and the result is shown in table 3. N-Desmethyl-2-carboxy OLZ was found to be the major metabolite in urine accounting for 36% of the radioactivity or 17% of the dose.

Discussion

Orally administered OLZ was well absorbed in dogs (absolute bioavailability: 73%) and to the extent of at least 55% in monkeys and
32% in mice, as demonstrated by the radioactivity recovered in urine. The elimination half-life of OLZ was relatively short in mice and monkeys (~3 hr) and long in dogs (~9 hr). The half-life of the drug in animals was much shorter than that reported for humans (~27 hr). The longer half-life for radioactivity in the three animals (figs. 2–4) indicated the formation of long-lived metabolites in these species. In mice and dogs radioactivity was predominantly eliminated in feces, but, in monkeys, as with humans (6), the major route of elimination of radioactivity was urine. For the present study, male mice, female dogs, and male and female monkeys were used. This was not a concern, because there were no sex-related differences in the plasma concentrations of OLZ. Also, OLZ pharmacokinetics were found not to be dose-dependent over the range used in this study.2

OLZ was subject to substantial first-pass metabolism, as evidenced by the fraction of the plasma AUC accounted for by OLZ compared with total radioactivity. At the t_{max}, OLZ accounted for 19%, 18%, and 8% of the radioactivity, in mice, dogs, and monkeys, respectively. A similar relationship was evident on the basis of AUC comparisons also. OLZ accounted for 10%, 14%, and 4% of the plasma AUC, respectively, in mice, dogs, and monkeys, indicating that OLZ was metabolized most extensively by the monkey and least by the dog among the three species. This observation is also reflected in the elimination half-life of the compound in these species (table 2). The ratio of AUC OLZ to AUC radioactivity found for the dog was the same as that reported for humans (6). Although many metabolites were present in plasma, attesting to the varied and extensive metabolism, the parent compound was the largest single component in plasma of all species studied, including humans.

Of the dose administered to mice, ~47% was excreted in the feces, whereas ~25% appeared in the urine within 24 hr of dosing. The principal urinary metabolites were 7-hydroxy OLZ glucuronide (fig. 10), 2-hydroxymethyl OLZ, and 2-carboxy OLZ accounting for ~10%, 4%, and 2% of the dose, respectively. Metabolites that were present in urine in lesser amounts were 7-hydroxy OLZ, N-desmethyl OLZ, and N-desmethyl-2-hydroxymethyl OLZ. In mouse plasma, mass spectral evidence was obtained for the presence of two isomeric glutathione conjugates of OLZ, although the corresponding N-acetyl-cysteine or cysteine conjugates were not detected in urine. Thus, on the basis of urinary metabolites, the principal metabolic pathways of OLZ in mice were aromatic hydroxylation, allylic oxidation (2-alkyl oxidation), and N-demethylation.

In dog urine (urinary excretion accounted for ~38% of the dose), the major urinary metabolite accounting for ~8% of the dose was 7-hydroxy-N-oxide OLZ (fig. 10). Other metabolites identified in urine were 2-hydroxymethyl OLZ (~3% of the dose), 2-carboxy OLZ (~3% of the dose), N-oxide OLZ (~1% of the dose), 7-hydroxy OLZ glucuronide (~1% of the dose), and N-desmethyl OLZ (<1% of the dose). The cysteine adducts of N-oxide OLZ and OLZ were also tentatively identified in urine extracts and were estimated to represent 2% and 1% of the dose, respectively. In dogs, therefore, OLZ underwent metabolic transformations mainly via aromatic hydroxylation, N-oxidation and 2-alkyl oxidation. The detection of the putative N-acetyl-cysteine conjugates implied the possible formation of their precursor glutathione conjugates. 2-Hydroxymethyl, N-oxide, N-desmethyl, 7-hydroxy OLZ, and its glucuronide were detected as circulating metabolites.

Approximately 55% of an oral dose given to monkeys was excreted in urine over a 7-day period, with ~48% appearing within the first 24 hr. The major urinary metabolite accounting for ~17% of the dose was a product of multiple oxidation, N-desmethyl-2-carboxy OLZ (fig. 10). Other metabolites identified in urine were N-oxide-2-hydroxymethyl OLZ (~6% of the dose), N-oxide-2-carboxy OLZ (~4% of the dose), N-desmethyl-2-hydroxymethyl OLZ (~4% of the dose), 2-carboxy OLZ (~3% of the dose), 2-hydroxymethyl OLZ (~3% of the dose), the N-acetylcysteine conjugate of OLZ (~1% of the dose), and 2-hydroxymethyl OLZ glucuronide (~1% of the dose). Thus, OLZ was biotransformed by the monkey mainly through double oxidation reactions involving the allylic carbon and the methyl piperazine nitrogen. Metabolism in the monkey was so efficient that intermediary metabolites such as N-oxide and N-desmethyl OLZ were not detected. Metabolism in the monkey was entirely driven by the allylic carbon attached to the thiophene ring, such that no oxidative metabolite was identified in which the methyl thiopehine moiety was not modified.

Thus, in mice and dogs, OLZ was metabolized through aromatic hydroxylation (forming phenolic metabolites and/or their glucuronide conjugates), allylic (alkyl) oxidation, thiol conjugation, and N-oxidation reactions (fig. 10). However, there were substantial species differences in the biotransformation of OLZ. The N-oxidation pathway was absent in mice. The 7-hydroxy-N-oxide metabolite, which is a product of aromatic and N-oxidation reactions, was formed only in the dog. The monkey was studied in efforts to get an animal model that produced OLZ 10-N-glucuronide (a major metabolite in humans). Although the monkey did not seem to form 10-N-glucuronide, the pattern of oxidative metabolism was similar to that of humans. In contrast to the other species studied, including humans, no intact OLZ was excreted in monkey urine. Among the four animal species studied, the monkey was unique in apparently not forming metabolites resulting from the oxidative attack of the benzene ring of OLZ.

Similarities in the metabolic fate of OLZ in animals and humans include the 2-alkyl hydroxylation, N-dealkylation, and N-oxidation pathways. Notable differences were that, direct glucuronidation, producing mainly the 10-N-glucuronide and to a lesser extent 4’-N-glucuronide, was the principal metabolic pathway in humans (6). These N-glucuronides were absent in animal samples, with the exception of a trace amount of 10-N-glucuronide in dog urine. Also, aromatic hydroxylation that was found to be a principal determinant of the clearance of the drug in mice and dogs did not seem to be an important pathway in humans.

In summary, orally administered OLZ was well absorbed and extensively metabolized by all species studied. Greater than 20 metabolites were identified in the four species studied, with the major metabolite in urine being strictly species dependent.

Acknowledgments. We thank Larry Spangle for the NMR data of 7-hydroxy-N-oxide OLZ.

References
2. R. Franklin, unpublished observations.
4. C. M. Beasley, T. Sanger, W. Satterlee, G. Tollefson, P. Tran, and S.


