EFFECTS OF PROTEIN CALORIE MALNUTRITION ON THE PHARMACOKINETICS OF KETAMINE IN RATS

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Received November 10, 2003; accepted April 15, 2004

This article is available online at http://dmd.aspetjournals.org

ABSTRACT:

The effect of protein calorie malnutrition (PCM) on the pharmacokinetics of ketamine (KET) enantiomers has been investigated. Six control and six PCM rats were administered 85 mg/kg racemic KET by intramuscular injection, and plasma concentrations of (S)- and (R)-KET, norketamine (NKET), and 5,6-dehydronorketamine (DNK) were measured using enantioselective gas chromatography. Pharmacokinetic profiles were analyzed using standard noncompartmental and compartmental modeling methods. The volumes of distribution were similar between control and PCM rats for (S)- and (R)-KET. However, total clearance of both KET enantiomers was decreased, resulting in an increase in systemic exposure (p < 0.05). The KET absorption rate was also increased in PCM rats. A decrease in the clearance of both NKET enantiomers led to a significant increase in exposure in PCM rats (p < 0.005), and modeling results could not exclude the possibility that PCM induced an increase in the fraction of KET following the NKET pathway, which may further contribute to this increase in exposure. An increase in exposure to DNK enantiomers was also evident in PCM animals compared with controls [p < 0.005 (DNK1); N.S. (DNK2)], which was in concordance with the decrease in apparent clearance values. These results show that PCM significantly alters the pharmacokinetics of KET and several of its metabolites.

Protein calorie malnutrition (PCM) is found in developed as well as underdeveloped countries (Torrín and Chew, 1994), and is the state reached when the body’s need for protein and energy fuels is not satisfied by the diet. PCM can result simply from inadequate food intake or can be the product of a disease state. In the latter case, diseases such as anorexia nervosa, cancer, and AIDS often lead to low food intake, inadequate nutrient absorption/utilization, or increased requirement of, or exaggerated loss of, nutrients. Approximately 50% of cancer patients will become malnourished during the course of their disease (Greene, 1988), and this can lead to impaired tolerance to chemotherapy and impaired immune function (Reilly et al., 1999). PCM also occurs in up to 65% of elderly hospitalized patients (Morley, 1998; Sullivan et al., 1999).

PCM has been shown to alter drug metabolism and disposition in children and in animal models (Parke and Ioannides, 1981; Anderson, 1988). In children, PCM reduced the clearance of antipyrine (Narang et al., 1977; Buchanan et al., 1979a), isoniazid, and acetaminophen (Jung, 1985), and altered oxidative drug metabolism (Campbell and Hayes, 1974; Fernandez et al., 1996; Walter-Sack and Klotz, 1996; Zhang et al., 1999), conjugation (Zhang et al., 1999), and protein binding (Varma, 1979). PCM induced alterations in drug metabolism and disposition appear to be multifactorial processes (Anderson, 1988; Zhang et al., 1999), and a complete understanding of the effect of nutritional status on drug pharmacokinetics would be highly desirable for effective clinical treatment.

Ketamine (KET; Fig. 1) is a chiral molecule that is often marketed as a racemic mixture [i.e., a 50:50 mixture of (S)-KET and (R)-KET enantiomers]. KET is a dissociative anesthetic agent, and the (S)- and (R)-enantiomers have significantly different pharmacokinetic and pharmacodynamic properties (White et al., 1982; Reich and Silvay, 1989). For example, the therapeutic (hypnotic, ataxic, and analgesic) potency of (S)-KET is between 2 and 4 times greater than that of the (R)-enantiomer (Marietta et al., 1977; White et al., 1985). Conversely, the posthypnotic stimulatory properties and agitated behavior that have limited the clinical use of KET are more common with (R)-KET (Marietta et al., 1977; White et al., 1982, 1985).

ABBREVIATIONS: PCM, protein calorie malnutrition; KET, ketamine; NKET, norketamine; DNK, dehydronorketamine; BrKET, bromoketamine; TFAA, trifluoroacetic anhydride; GC, gas chromatography; PH, peak height; PHR, peak height ratio; PK, pharmacokinetic; V, volume of distribution; ka, first-order absorption rate constant; kely, first-order elimination rate constant; CI, clearance; t1/2, half-life; T max, time to maximal concentration; Cmax, maximal concentration; AUC, area under the concentration-time curve.
KETAMINE PHARMACOKINETICS IN PROTEIN-DEFICIENT RATS

have not been established. The enzymes that mediate the hydroxylation pathways also have not been identified.

During the course of a study on the effect of PCM on the metabolism of the anti-cancer agent ifosfamide (Granvil et al., 1994), it was observed that KET immobilized PCM rats for a longer period of time than control rats. The purpose of this study is to investigate the effects of PCM on the pharmacokinetics of (R)- and (S)-KET, as well as the enantiomers of NKET and 5,6-DNK.

Materials and Methods

Chemical and Biological Reagents. Racemic and individual enantiomers of KET and NKET, along with 5,6-DNK and bromoketamine ([R,S]-BrKET), were generous gifts from Pfizer Global Research and Development (Groton, CT). Trifluoroacetic acid anhydride (TFAA) was purchased from Pierce Chemical (Rockford, IL), and both triethylamine and 2,2-dimethoxypropane were purchased from Sigma-Aldrich (St. Louis, MO). Ethanol and tolune ("Baker Analyzed") were obtained from J. T. Baker (Phillipsburg, NJ). Rat plasma for assay validation and calibration purposes was obtained from Pel Freez (Rogers, AR).

Animals and Treatment Protocol. This study was carried out in accordance with a protocol approved by the McGill University Animal Care Committee. Male Sprague-Dawley rats (specific pathogen free), 39 to 42 days old with an initial weight of 150 to 175 g, were obtained from Charles River Canada (Montreal, PQ, Canada). Rats were housed in a conventional facility with a 12-h light/dark cycle. After 1 to 3 days acclimation, rats started to receive isocaloric diets containing either 22.5% (control) or 5.5% (deficient) protein, as described by Varma (1979). The control diet (Rat Chow 5012) was purchased from Purina (St. Louis, MO) and the low protein diet (TD 77210) from Harlan (Indianapolis, IN). All animals were fed the control or deficient diets ad libitum for 17 to 20 days before KET administration and blood sampling. On the 15th day, plasma protein levels for control rats were, on average, 5.69 g of protein per 100 ml of plasma (range 5.32–6.01) and for PCM rats were 4.34 g/100 ml (range 4.01–4.94). Immediately before KET administration (at an age of 56–63 days), control rats weighed, on average, 345 g (range 327–363 g) and PCM rats, 199 g (range 186–208 g). Ketamine hydrochloride [100 mg of ketamine base/ml of aqueous solution, pH 3.5–5.5; Ketaset; Wyeth-Ayerst (Princeton, NJ)] was administered intramuscularly (thigh muscle) at a dose of 85 mg/kg. Blood sampling took place for up to and including 180 min post-KET dosing. Six of the rats (three control and three PCM) had samples taken at 10, 40, 90, 120, and 150 min postdose, whereas the other six rats (three control and three PCM) had blood samples removed at 20, 50, 90, 120, and 180 min postdose. Blood samples (approximately 0.3 ml each) were collected with a 21 to 23 gauge needle from the tail vein into heparinized tubes, and the samples were kept on ice until centrifugation (5 min at 10,000 C for 15 min, the reaction mixture was cooled and then shaken with 1 ml of 5% sodium bicarbonate solution for 5 min. Subsequent to centrifugation, the tolune layer was removed and dried down under a steady nitrogen flow in the presence of 20 µl of propan-2-ol, and the residue was reconstituted with 50 µl of ethanol, awaiting analysis by GC.

Sample Preparation. Trifluoroacetic acid anhydride was used for the preparation of trifluoracetyl derivatives of KET, NKET, and 5,6-DNK before analysis by enantioselective gas chromatography (GC). The plasma concentrations of the derivatized KET, NKET, and 5,6-DNK were determined using a validated enantioselective GC assay on a Hewlett Packard (Palo Alto, CA) 5890 series II GC with electron capture detection, equipped with an HP7673 autosampler, operating in the splitless mode. The column contained a cyclodextrin-based
chiral stationary phase [ChiralDEX GTA capillary column, 20 m, 0.25-mm i.d.; Advanced Separation Technologies Inc. (Astec), Whippany, NJ], and the carrier gas was helium at a flow rate of 1.7 ml/min. The injector and detector temperatures were 250 and 300°C, and 0.5 μl of each sample in ethanol was injected onto the gas chromatograph. The total duration of each run was 55 min, involving a series of temperature gradients. At zero time, the temperature was 40°C, ramping up to 140°C following sample injection at a rate of 20°C /min. After holding at 140°C for 10 min, the oven was taken from 140°C to 150°C at a rate of 0.5°C/min and held at 150°C for 14 min. After this time, a temperature of 170°C was achieved at a rate of increase of 10°C/min which was held for 4 min, resulting in a total run time of 55 min.

Due to high levels of KET and, particularly, NKET enantiomers in rat plasma, standard curves were taken up to 20 g/ml. Inter- and intraday coefficients of variation were not satisfactory (4% for (R)- and (S)-KET, NKET, and DNK enantiomers were fitted sequentially to one-compartment linear mammillary plasma clearance models (Fig. 2). The (R)- and (S)-enantiomer pathways were modeled separately. The rate of change of KET concentrations in the central compartment (C<sub>KET</sub>) was described by the following equation:

\[
\frac{dC_{KET}}{dt} = k_a \cdot \text{Dose} \cdot e^{-k_t \cdot t} - k_{el} \cdot C_{KET}
\]

where \(k_a\) is a first-order absorption rate constant, \(k_t\) is a first-order elimination rate constant, and \(V_{KET}\) is the volume of distribution of ketamine. The fraction of KET absorbed into the bloodstream following i.m. injection was assumed to be complete (Clements et al., 1982), and total systemic clearance was secondarily calculated, where \(C = k_a \cdot V_{KET}\).

The pharmacokinetics of KET enantiomers were fixed according to eq. 1 and the resulting estimated parameters, and the concentrations of NKET (C<sub>NKET</sub>) were characterized subsequently using the following equation:

\[
\frac{dC_{NKET}}{dt} = k \cdot C_{NKET} - k_{em} \cdot C_{NKET}
\]

where \(k\) and \(k_{em}\) are first-order rate constants of formation and elimination. The concentrations of DNK (C<sub>DNK</sub>) were modeled in a similar fashion, such that

\[
\frac{dC_{DNK}}{dt} = k_2 \cdot C_{DNK} - k_{em2} \cdot C_{DNK}
\]

where \(k_2\) and \(k_{em2}\) are first-order rate constants of formation and elimination of DNK enantiomers and C<sub>NKET</sub> is fixed according to eq. 2. Authentic standards of (S)- and (R)-DNK were not available during GC analysis; thus, the first enantiomer to elute from the chiral column was named DNK1 and the second DNK2 (Fig. 3B). However, initial modeling efforts gave evidence to suggest that DNK1 was (R)-DNK and DNK2 was (S)-DNK, inasmuch as a fit could be achieved only when DNK1 was fitted with (R)-KET and (R)-NKET, and DNK2 with the (S)-enantiomer pathway and not vice versa. The total AUC values of the fitted profiles for KET and DNK enantiomers were calculated from the estimated parameters according to

\[
\text{AUC} = \lim_{t \to \infty} (C)
\]

where \(C\) is C<sub>NKET</sub> or C<sub>DNK</sub> in the Laplace domain. Thus,

\[
\text{AUC}_{\text{N,KET}} = k \cdot \text{Dose} \cdot \left( \frac{k \cdot k_t \cdot V}{(s + k_t)(s + k_a)(s + k_{em})} \right) = \frac{k \cdot \text{Dose}}{V \cdot k_a \cdot k_{em}}
\]

and

\[
\text{AUC}_{\text{DNK}} = k_2 \cdot k \cdot k_{em} \cdot \left( \frac{k_2 \cdot k_t \cdot V}{(s + k_2)(s + k_a)(s + k_{em})(s + k_{em2})} \right) = \frac{k_2 \cdot k \cdot \text{Dose}}{V \cdot k_a \cdot k_{em} \cdot k_{em2}}
\]
Hence, the apparent clearance \((Cl/F)\) of the metabolites was calculated as Dose/AUC, and the apparent volumes of distribution \((V/F)\) were calculated as the ratio of the \(Cl/F\) terms to their respective elimination rate constant \((k_{\text{em}}\) or \(k_{\text{em2}}\)). In this manner, the \(F\) term for NKET is \(f_r\), and \((f_r \times f_{r2})\) for DNK (Fig. 2).

Concentration data were analyzed using a naive-pooling approach, and all model parameters were estimated using WinNonlin. Nonlinear regression analysis was conducted with data weighted by \(1/Y\), and goodness-of-fit was assessed by system convergence, Akaike information criterion, correlation coefficients \((r^2)\), examination of residuals, and visual inspection.

**Results**

This study comprised, in total, six control and six PCM rats, all receiving an 85 mg/kg intramuscular dose of KET hydrochloride. PCM rats were under the anesthetic effects of KET for considerably longer than control rats. For five of the control rats, the duration of immobilization was between 60 and 90 min, with one of the control rats not undergoing complete anesthesia at all, being hypoactive but still mobile. In contrast, whereas one PCM rat started to move approximately 60 min after KET.
dosing, the remaining five were immobilized for ≥90 min (90, 105, 115, >120, >120 min).

Under the analytical conditions, the enantioselective separation of the enantiomers of KET, NKET, DNK, and BrKET was achieved in a single run without significant interference from the plasma matrix. Representative chromatograms from blank plasma and from 120 min postadministration are presented in Fig. 3, A and B. The retention times of the enantiomers of KET were 36.9 (R) and 37.6 (S) min; of NKET, 19.9 (S) and 21.9 (R) min; of 5,6-DNK, 29.6 and 35.0 min (R- and S-enantiomers not defined); and of BrKET, 47.4 and 48.1 min (R- and S-enantiomers not defined). The enantioselective separation values (ε₆₃) were 1.02, 1.11, 1.19, and 1.02, respectively. In addition to peaks corresponding to enantiomers of KET, NKET, DNK, and BrKET, peaks were observed at 21.5, 26.5, and 31.0 min that increased with sampling time and were not present in blanks (Fig. 3B). However, the compounds yielding these peaks were not characterized.

The net exposure to (S)-KET, (R)-KET, (S)-NKET, (R)-NKET, and DNK1 was significantly increased due to PCM. Although exposure to DNK2 was increased, the difference between control and PCM rats was not statistically significant (Table 1). For NKET enantiomers, the exposure was increased approximately 4-fold (p < 0.005) and for DNK enantiomers, the AUC was almost doubled. PCM did not affect peak concentrations and the time to reach them for both ketamni enantiomers (Table 1). Using noncompartmental analysis, PCM decreased the total median systemic clearance of (S)-KET [5.81 (2.60–6.29) (PCM) versus 8.75 (6.03–15.55) l h⁻¹ kg⁻¹ (control); p < 0.005] and (R)-KET [5.66 (3.08–6.65) (PCM) versus 7.9 (5.44–13.69) l h⁻¹ kg⁻¹ (control); p < 0.05]. The terminal half-lives of (S)- and (R)-KET were increased by 83% and 50% (Table 1). There was a trend toward higher T₉⁰ values for NKET enantiomers in PCM rats, particularly for (S)-NKET, although this was not statistically significant. Peak (S) and (R)-NKET levels and terminal t₁/₂ values were approximately 3 times higher (Table 1). A large delay in T₉⁰ values was shown for DNK enantiomers due to PCM (p < 0.005), and peak concentrations were higher for both enantiomers, although this was statistically significant for DNK1 only (Table 1).

Considering the control and PCM rat groups individually, no enantio-specific differences were apparent with respect to (S)- and (R)-KET concentration profiles. On the other hand, peak (S)-NKET concentrations were higher than those of (R)-NKET within both control and PCM rats, reflected in significantly greater AUC and C₉⁰ values for the (S)-enantiomer. Peak DNK1 concentrations and AUC values were also consistently higher than those of DNK2 within the control and PCM groups.

The final models used for the kinetic analysis (Fig. 2) suitably described the PC-induced alterations in KET and NKET pharmacokinetics, fitting the plasma concentrations of these compounds well (Fig. 4). Although the chosen model provided a description of DNK data from both rat groups, the precision of the parameter estimates was low for the control rat group (%CV >50%), leading to some uncertainty in interpretation of the data.

The volume of distribution of KET did not appear to be markedly altered due to PCM (Table 2). However, the absorption rate constant was more than doubled for (S)-KET and increased by approximately 60% for (R)-KET, whereas the elimination rate constant and total clearance were reduced by 46% and 32% for (S)-KET and 33% and 27% for (R)-KET. For both NKET enantiomers, the formation rate constant (k) was increased, whereas Vₙ⁰ₙ/KETF was decreased due to PCM. The elimination rate constant for (S)- and (R)-NKET was only slightly reduced due to PCM (Table 2), although the apparent clearance of both NKET enantiomers was decreased approximately 4-fold. The clearance of both DNK1 and DNK2 was also decreased (Table 2). Differences in the DNK modeling parameters must be evaluated cautiously in light of the relatively low precision of the estimates. For both enantiomers, the formation rate constant appeared to be decreased, whereas Vₙ⁰ₙ/KETF was increased due to PCM.

Within control and PCM rat groups, there did not appear to be any relevant differences between (S)- and (R)-enantiomers for any of the estimated KET parameters. Likewise, there were no striking differences between (S)- and (R)-NKET for the k or Vₙ⁰ₙ/KETF terms. However, the apparent clearance and elimination rate constant of (R)-NKET were approximately 1.5 to 2 times higher than for the (S)-enantiomer. The apparent clearance was slightly higher for DNK2 than for DNK1 as well.

**Discussion**

This is the first study in which the pharmacokinetics of (S)- and (R)-KET in rats, including metabolites, are described, with the aim of

**TABLE 1**

**Noncompartmental analysis of ketamine and metabolite enantiomers in control and protein calorie malnourished rats**

<table>
<thead>
<tr>
<th>Parameters</th>
<th>(S)-Ketamine</th>
<th>(R)-Ketamine</th>
<th>(S)-Norketamine</th>
<th>(R)-Norketamine</th>
<th>Dehydronorketamine 1</th>
<th>Dehydronorketamine 2</th>
</tr>
</thead>
<tbody>
<tr>
<td>Control</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>C₉⁰ (µg·ml⁻¹)</td>
<td>4.81</td>
<td>5.00</td>
<td>3.26</td>
<td>2.08†</td>
<td>1.49</td>
<td>1.04</td>
</tr>
<tr>
<td>(2.52–5.88)</td>
<td>(2.46–6.00)</td>
<td>(2.02–3.52)</td>
<td>(1.19–2.68)</td>
<td>(1.15–1.65)</td>
<td>(0.22–1.98)</td>
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</tr>
<tr>
<td>T₉⁰ (h)</td>
<td>0.33</td>
<td>0.33</td>
<td>0.81</td>
<td>0.81</td>
<td>0.81</td>
<td>0.81</td>
</tr>
<tr>
<td>(0.17–0.67)</td>
<td>(0.17–0.67)</td>
<td>(0.66–1.50)</td>
<td>(0.66–0.83)</td>
<td>(0.67–0.83)</td>
<td>(0.67–0.83)</td>
<td></td>
</tr>
<tr>
<td>AUC (mg·h·l⁻¹)</td>
<td>4.85</td>
<td>5.38</td>
<td>4.88</td>
<td>3.11</td>
<td>2.07</td>
<td>1.47</td>
</tr>
<tr>
<td>(2.73–7.03)</td>
<td>(3.11–7.83)</td>
<td>(3.45–6.21)</td>
<td>(1.73–3.90)</td>
<td>(1.43–2.29)</td>
<td>(0.29–2.78)</td>
<td></td>
</tr>
<tr>
<td>T₁/₂ (h)</td>
<td>0.54</td>
<td>0.55</td>
<td>0.76</td>
<td>0.55</td>
<td>0.51</td>
<td>0.56</td>
</tr>
<tr>
<td>(0.33–0.71)</td>
<td>(0.31–0.79)</td>
<td>(0.31–0.79)</td>
<td>(0.31–0.79)</td>
<td>(0.31–0.79)</td>
<td>(0.31–0.79)</td>
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</tr>
<tr>
<td>PCM</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>C₉⁰ (µg·ml⁻¹)</td>
<td>5.08</td>
<td>5.04</td>
<td>8.51**</td>
<td>5.67**</td>
<td>1.97**</td>
<td>1.44†</td>
</tr>
<tr>
<td>(2.94–7.04)</td>
<td>(2.94–6.46)</td>
<td>(7.08–11.31)</td>
<td>(4.86–6.64)</td>
<td>(1.70–2.50)</td>
<td>(0.75–1.85)</td>
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</tr>
<tr>
<td>T₉⁰ (h)</td>
<td>0.33</td>
<td>0.33</td>
<td>1.17</td>
<td>0.83</td>
<td>2.00</td>
<td>2.00**</td>
</tr>
<tr>
<td>(0.17–0.90)</td>
<td>(0.17–0.90)</td>
<td>(0.78–2.00)</td>
<td>(0.78–1.50)</td>
<td>(1.50–2.50)</td>
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</tr>
<tr>
<td>AUC (mg·h·l⁻¹)</td>
<td>7.32*</td>
<td>7.53*</td>
<td>16.88**</td>
<td>11.16**</td>
<td>3.70</td>
<td>2.87</td>
</tr>
<tr>
<td>t₁/₂ (h)</td>
<td>0.99**</td>
<td>0.82*</td>
<td>2.29</td>
<td>2.01</td>
<td>2.01</td>
<td>2.01**</td>
</tr>
<tr>
<td>(0.73–2.48)</td>
<td>(0.68–2.00)</td>
<td>(1.34–3.12)</td>
<td>(1.34–3.12)</td>
<td>(1.34–3.12)</td>
<td>(1.34–3.12)</td>
<td></td>
</tr>
</tbody>
</table>

° Calculated from 0 to infinity for KET, and to 3 h for NKET and DNK.
† Calculated using pooled concentration-time data.
‡ Significantly different between control and PCM rats (p < 0.05).
‡† Significantly different between control and PCM rats (p < 0.005).
†† Significantly different between enantiomers (p < 0.05).
‡‡ Significantly different between enantiomers (p < 0.01).
elucidating the effects of PCM on the pharmacokinetics of KET. Although the number of animals in the study was small, this descriptive study suggests that PCM not only leads to a reduction in the clearance of both (S)- and (R)-KET, but also affects the pharmacokinetics of several KET metabolites by altering the clearance and quite possibly formation mechanisms.

Table 2
Pharmacokinetic modeling parameters for ketamine and metabolite enantiomers in control and protein calorie malnourished rats

<table>
<thead>
<tr>
<th>Parameters</th>
<th>(S)-Ketamine</th>
<th>(R)-Ketamine</th>
<th>(S)-Norketamine</th>
<th>(R)-Norketamine</th>
<th>Dehydronorketamine 1</th>
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<tbody>
<tr>
<td>Control</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>$k$ (min$^{-1}$)$^{a}$</td>
<td>0.0976 (34)</td>
<td>0.0964 (35)</td>
<td>0.0332 (11)</td>
<td>0.0279 (15)</td>
<td>0.375 (72)</td>
<td>0.194 (166)</td>
</tr>
<tr>
<td>$k_e$ (min$^{-1}$)$^{b}$</td>
<td>0.0202 (15)</td>
<td>0.0189 (15)</td>
<td>0.0297 (11)</td>
<td>0.0495 (15)</td>
<td>0.518 (73)</td>
<td>0.550 (167)</td>
</tr>
<tr>
<td>$Cl/F$ (l $\cdot$ h$^{-1}$ $\cdot$ kg$^{-1}$)$^{c}$</td>
<td>7.39 (7.1)</td>
<td>6.84 (7.2)</td>
<td>6.60 (5.2)</td>
<td>12.1 (6.2)</td>
<td>16.8 (5.5)</td>
<td>18.7 (11)</td>
</tr>
<tr>
<td>$V/F$ (l $\cdot$ kg$^{-1}$)$^{c}$</td>
<td>6.08 (15)</td>
<td>6.04 (15)</td>
<td>3.70 (11)</td>
<td>4.09 (15)</td>
<td>0.540 (72)</td>
<td>0.566 (166)</td>
</tr>
<tr>
<td>PCM</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>$k$ (min$^{-1}$)$^{a}$</td>
<td>0.195 (54)</td>
<td>0.153 (38)</td>
<td>0.0746 (13)</td>
<td>0.0676 (15)</td>
<td>0.0134 (28)</td>
<td>0.0171 (44)</td>
</tr>
<tr>
<td>$k_e$ (min$^{-1}$)$^{b}$</td>
<td>0.0314 (14)</td>
<td>0.0127 (12)</td>
<td>0.0235 (16)</td>
<td>0.0349 (16)</td>
<td>0.0398 (31)</td>
<td>0.116 (45)</td>
</tr>
<tr>
<td>$Cl/F$ (l $\cdot$ h$^{-1}$ $\cdot$ kg$^{-1}$)$^{c}$</td>
<td>5.05 (7.2)</td>
<td>5.00 (6.3)</td>
<td>1.59 (6.7)</td>
<td>2.58 (6.3)</td>
<td>7.66 (9.6)</td>
<td>10.8 (7.6)</td>
</tr>
<tr>
<td>$V/F$ (l $\cdot$ kg$^{-1}$)$^{c}$</td>
<td>6.80 (11)</td>
<td>6.56 (11)</td>
<td>1.13 (13)</td>
<td>1.23 (15)</td>
<td>3.21 (28)</td>
<td>1.55 (43)</td>
</tr>
</tbody>
</table>

$a$ Represents $k_a$, $k_1$, and $k_2$ for KET, NKT, and DNK, respectively.

$b$ Represents $k_{el}$, $k_{em}$, and $k_{em2}$ for KET, NKT, and DNK, respectively.

$c$ $F$ = 1 (KET), $fr$ (NKET), and $fr1$ $fr2$ (DNK).

Fig. 4. Concentration-time profiles of ketamine (first row), norketamine (second row), and 5,6-dehydronorketamine (third row) enantiomers following ketamine administration to control and PCM rats. The filled circles represent the observed concentration values. The lines through the data depict the model-fitted profiles.
Following both noncompartmental analysis and fitting of concentration data from control rats to the selected pharmacokinetic model, it appeared that there were no significant differences in the volume of distribution, clearance, half-life or AUC between (R)- and (S)-KET. A lack of enantiomeric specificity in the exposure to KET is in accordance with previous findings in rats in which KET plasma levels were determined after i.v. administration of the separate KET enantiomers (Marietta et al., 1977; Ryder et al., 1978). A value of approximately 6 l/kg for the volume of distribution of KET enantiomers (Table 2) was the same whether determined using modeling techniques or noncompartmental analysis (data not shown). However, the volume of distribution of racemic KET reported in the literature for species other than rat is approximately 3 l/kg (Kaka and Hayton, 1980; Domino et al., 1982, and references therein). A species difference in plasma protein or tissue binding could lead to such an effect, although KET has been found to exhibit very low protein binding (Duvaldestin, 1981; Wagner and O’Hara, 1997).

Upon pharmacokinetic analysis of concentration data from rats with PCM, the absorption of KET into the systemic circulation after i.m. injection appeared to be enhanced compared with that seen in control rats (Table 2). The increase in \(k_e\) for both KET enantiomers is most likely due to the smaller muscle mass of PCM animals (Table 2). Protein deficiency, and the loss of lean body mass that accompanies it, did not seem to affect the volume of distribution of KET. However, a reduction in clearance of both KET enantiomers is probably the major cause of the increased systemic exposure and half-life of (R)- and (S)-KET observed in PCM animals (Fig. 4; Table 1), potentially contributing to the increased anesthetic effects of KET in this rat group. Studies illustrating a PCM-related inhibition of P450s in rodents have been previously reported (Fernandez et al., 1996; Zhang et al., 1999). This inhibition includes the rodent counterparts of those reportedly associated with KET N-demethylation in humans (CYP2B, CYP3A, and CYP2C) (Yanagihara et al., 2001; Hijazi and Bouliou, 2002). It follows that the apparent reduction in the overall clearance of KET in PCM rats would occur as a result of such a reduction in phase I metabolism.

N-Demethylation is considered the major metabolic pathway for both (S)- and (R)-KET in rat and in humans (Wieber et al., 1975; Marietta et al., 1977; Adams et al., 1981; Trevor et al., 1983). NKET enantiomers are active metabolites of KET and, despite having a shorter duration of action at equivalent doses (Cohen and Trevor, 1974), their pattern of activity follows that of their parents, such that the (S)-enantiomer has the greater anesthetic properties, whereas (R)-NKET appears to cause more pronounced posthypnotic stimulation (Hong and Davission, 1982). Within the control rat group, the net exposure to (S)-NKET was 57% higher than that of (R)-NKET (Table 1), which is consistent with the work of Ryder et al. (1978). Although (S)- and (R)-NKET formation and apparent volume of distribution (\(k_d\) and \(V_{NKET}/F\)) appeared to be similar, the elimination rate constant (\(k_{em}\)) and apparent clearance for (R)-NKET (\(CL/F\)) were double the estimation for (S)-NKET (Table 2), offering an explanation for the observed higher plasma levels of (S)-NKET (Fig. 4).

PCM led to a 4-fold increase in systemic exposure to both NKET enantiomers and an equivalent increase in half-life compared with controls (Table 1). The median AUC ratios of NKET to KET in PCM rats were more than double those observed for control animals [2.31 versus 1.01 and 1.48 versus 0.58 for (S)- and (R)-enantiomers]. These changes are most likely a result of a decrease in the clearance of (R)- and (S)-NKET due to PCM (\(CL/F\) decreased approximately 4-fold) (Table 2). However, it is possible that an increase in the fraction of (R)- and (S)-KET following the N-demethylation pathway also occurs. Assuming that PCM does not alter NKET volume of distribution, the decrease in \(V_{NKET}/F\) might indicate an increase in the fraction (\(fr\)) of (S)- and (R)-KET following the N-demethylation pathway in PCM animals. One possible explanation for this phenomenon could be an inhibition of the alternative metabolic pathways of KET (i.e., 4-, 5-, or 6-hydroxylation; Fig. 1), leaving more KET available to N-demethylation. The concept of “compensatory” increases in one metabolic pathway due to an inhibition of a competing pathway has previously been observed for glucuronidation versus sulfation (Morris and Pang, 1987, and references therein). The increase in NKET exposure and half-life suggests that plasma concentrations of NKET might remain above some minimum effective concentration for an extended period of time, thus contributing to the increased anesthetic effects of KET in PCM rats.

The other metabolite that was measured in this study, DNK, is proposed to arise from nonenzymatic dehydration of 5-hydroxy-NKET (Adams et al., 1981) and quite probably 6-hydroxy-NKET (6-OH-NKET), as an artifact of GC analytical procedures (Fig. 1). From in vitro studies, Trevor et al. (1983) indicate that only the (R)-enantiomer of 5-OH-NKET and the (S)-enantiomer of 6-OH-NKET are formed in rat liver. Thus, it would follow that the nonenzymatic formation of (R)-DNK should reflect and give some insight into the 5-hydroxylation of (R)-NKET and that of (S)-DNK should correspond to the 6-hydroxylation of (S)-NKET. Although, after GC analysis, chromatograms displayed two measurable DNK peaks (DNK1 and DNK2; Fig. 3B), their enantiomeric identity could not be ascertained due to the lack of separate authentic standards of (S)- and (R)-DNK. However, in an attempt to include DNK in the PK model, it was determined that DNK1 fitted only with the (R)-enantiomer pathway, thus possibly arising from (R)-5-OH-NKET, and DNK2 fitted only with the (S)-pathway, thus possibly arising from (S)-6-OH-NKET. The systemic exposure to DNK1 was approximately 40% higher than that to DNK2 within control rats and 30% higher within PCM rats (Table 1).

PCM led to an approximate 2-fold increase in systemic exposure of rats to both DNK enantiomers (Table 1; Fig. 4), and there was a lack of any obvious terminal phase by the end of the 3-h experiment. In concurrence with these findings, modeling results suggested that the clearance (\(CL/F\)) of both DNK enantiomers was halved in PCM animals compared with controls (Table 2). These results could reflect an inhibition in the clearance of (R)-5-OH- and (S)-6-OH-NKET due to PCM. Interestingly, median DNK/NKET AUC ratios in PCM rats were approximately half the values for control rats (0.33 versus 0.67 and 0.17 versus 0.30 for DNK1 and DNK2), which might suggest that there was a decrease in the fraction of NKET passing through the DNK pathway. This is supported indirectly by the increase in \(V_{DNK}/F\) terms calculated for PCM rats (Table 2). Since the \(F\) term for \(V_{DNK}\) is \(fr \cdot fr_2\), we hypothesize that the decrease in \(fr_2\) is more significant than any potential increase in \(fr\) (see \(V_{DNK}/F\) in Table 2), obviously further relying on the assumption that PCM does not affect \(V_{DNK}\). Taken together, the decreased clearance of DNK sufficiently offsets any decrease in DNK formation, resulting in augmented DNK exposure in PCM animals. The delay before maximal plasma levels of either DNK enantiomer were achieved in PCM rats, illustrated by \(T_{max}\) values (Table 1), could reflect a delay in the formation of both (R)-5-OH-NKET and (S)-6-OH-NKET from NKET. In an attempt to improve the fitting of DNK enantiomers to the model, an additional transit (intermediary) compartment between NKET and DNK was included, but this failed to improve model fitting criteria (data not shown).

In conclusion, it appears that PCM alters the pharmacokinetics of KET, leading to changes in absorption and elimination of both (R)- and (S)-KET. PCM also affects the pharmacokinetics of the KET.
metabolites, primarily via clearance mechanisms but possibly also by affecting the fraction of drug and intermediate metabolites following a particular metabolic pathway. These changes offer an explanation for the increase in immobilization time seen in PCM rats following KET administration, compared with controls. Although the specific P450s responsible for these PCM-related effects have not been identified, it can be suggested that this condition has the potential to directly impact the pharmacokinetics of many therapeutic agents that are metabolized by P450s.

Acknowledgments. We thank Advanced Separation Technologies Inc. (Astec), Whippanny, NJ, for expert help in the development of the enantioselective assay for ketamine and its metabolites.

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