Metabolism, Excretion, and Pharmacokinetics of [14C]Tigecycline, a First-In-Class Glycylcycline Antibiotic, after Intravenous Infusion to Healthy Male Subjects

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ABSTRACT:

Tigecycline, a novel, first-in-class glycylcycline antibiotic, has been approved for the treatment of complicated intra-abdominal infections and complicated skin and skin structure infections. The pharmacokinetics, metabolism, and excretion of [14C]tigecycline were examined in healthy male volunteers. Tigecycline has been shown to bind to bone; thus, to minimize the amount of radioactivity binding to bone and to maximize the recovery of radioactivity, tigecycline was administered intravenously (30-min infusion) as a single 100-mg dose, followed by six 50-mg doses, every 12 h, with the last dose being [14C]tigecycline (50 µCi). After the final dose, the pharmacokinetics of tigecycline in serum showed a long half-life (55.8 h) and a large volume of distribution (21.0 l/kg), whereas radioactivity in serum had a shorter half-life (6.9 h) and a smaller volume of distribution (3.3 l/kg). The major route of elimination was feces, containing 59% of the radioactive dose, whereas urine contained 32%. Unchanged tigecycline was the predominant drug-related compound in serum, urine, and feces. The major metabolic pathways identified were glucuronidation of tigecycline and amide hydrolysis followed by N-acetylation to form N-acetyl-9-amino-nocycline. The glucuronide metabolites accounted for 5 to 20% of serum radioactivity, and approximately 9% of the dose was excreted as glucuronide conjugates within 48 h. Concentrations of N-acetyl-9-amino-nocycline were approximately 6.5% and 11% of the tigecycline concentrations in serum and urine, respectively. Excretion of unchanged tigecycline into feces was the primary route of elimination, and the secondary elimination pathways were renal excretion of unchanged drug and metabolism to glucuronide conjugates and N-acetyl-9-amino-nocycline.

Tigecycline [GAR-936, (4S,4aS,5aR,12aS)-9-(2-tert-butylationino-acetylamino)-4,7-bis-dimethylamino-3,10,12,12a-tetrahydroxy-1,11-dioxo-1,4,4a,5,6a,11,12a-octahyronaphthacene-2-carboxamide] (Fig. 1), a novel, first-in-class glycylcycline and an analog of the semisynthetic antibiotic minocycline, is a potent, broad-spectrum antibiotic that acts by inhibition of protein translation in bacteria. Glycylcycline antibiotics, including tigecycline and others (N,N-di-methylglycylamido-9-aminonocycline and N,N-dimethylglycylamido-9-amino-6-demethyl-6-deoxytetracycline), are active in vitro against multiple antibiotic-resistant pathogenic bacteria, including methicillin-resistant Staphylococcus aureus and vancomycin-resistant enterococci (Fraishe et al., 1995; Weiss et al., 1995). Tigecycline is also active against bacterial strains carrying the two major forms of antibiotic resistance, active efflux and ribosomal protection (Schnapper and Hillen, 1996). In patients, intravenous tigecycline (50 mg, twice daily) was effective against complicated skin and skin-structure infections and complicated intra-abdominal infections with an acceptable safety profile (Murray et al., 2003; Postier et al., 2004).

The pharmacokinetic profile of tigecycline in healthy human subjects is characterized by a long half-life of 37 to 67 h and a large volume of distribution (Vdss) of 7 to 10 l/kg (Muralidharan et al., 2005). The metabolism of glycylcycline antibiotics has not previously been investigated in humans. However, other antibiotics containing the tetracycline ring structure, such as doxycycline, tetracycline, chlorotetracycline, and demethylchlorotetracycline, generally undergo little or no metabolism (Kelly and Buyske, 1960; Kelly et al., 1961; Eisner and Wulf, 1963; Swarz, 1976; Nelis and De Leeheer, 1981). Minocycline is an exception to this pattern, since it undergoes metabolism in humans via hydroxylation and N-demethylation (Nelis and De Leeheer, 1982; Böcker et al., 1991).

The present study was conducted to evaluate the pharmacokinetics, metabolic disposition, and mass balance of a 50-mg intravenous dose of 14C-labeled tigecycline in healthy male volunteers. [14C]Tigecycline has been shown to distribute extensively into bone following a single i.v. dose to rats, with bone to plasma ratios for radioactivity as high as 2000 (Tombs, 1999). Even with this extensive distribution into bone, the estimated exposure to radioactivity after a 50-µCi (50-mg) intravenous dose of [14C]tigecycline to humans was 0.023 rem or 0.46% of the maximum allowable exposure from radioactive drugs for

ABBREVIATIONS: LC-MS/MS, liquid chromatography-tandem mass spectrometry; HPLC, high performance liquid chromatography; LC/MS, liquid chromatography/mass spectrometry; AUC, area under the curve; AUCt, area under the concentration-time curve; HMBC, heteronuclear multiple bond correlation; MRT, mean residence time; CL, clearance.
human research subjects (5 rem). However, to minimize the amount of 14C-labeled material binding to bone and to maximize the recovery of radioactivity, the 14C-labeled tigecycline dose was administered after multiple doses (350 mg over 3 days) of unlabeled tigecycline. Using this dosing regimen, previous studies have shown that steady-state tigecycline serum concentrations were achieved on day 4 (Sun et al., 2005). Thus, for the current study, it was assumed that by the time of the [14C]tigecycline dose, steady-state or close to steady-state conditions had been achieved in serum, as well as in other tissues such as bone.

### Materials and Methods

#### Materials

Tigecycline, as a lyophilized powder in vials containing a 50-mg dose of drug, was provided by Wyeth Research (Collegeville, PA). The manufacturing and packaging of [14C]tigecycline were carried out at the Parenteral Medications Laboratories, College of Pharmacy, University of Tennessee Health Science Center (Memphis, TN). Nonradiolabeled tigecycline drug substance was used to dilute radiolabeled tigecycline to make the final 14C-drug substance. Radiolabeled tigecycline for injection was prepared in 5-ml clear glass vials, each containing 53 mg of lyophilized, sterilized free tigecycline powder. The final [14C]tigecycline drug product had a specific activity of 9.2 mCi/mmol (1.00 μCi/mg) with a radiochemical purity of 98.6% and a chemical purity of 99.2%. Appearance, strength, identity, specific activity, and purity testing of [14C]tigecycline for injection was performed by ABC Laboratories (Columbia, MO). Bacterial endotoxin and sterility testing of [14C]tigecycline for injection was carried out by Wyeth Research.

An additional batch of [14C]tigecycline (95.3 μCi/mg, 97.2% radiochemical purity), used in control samples, and [r-butyld-9]tigecycline, the internal standard for the determination of tigecycline in serum and urine, were received from GE Healthcare (Chalfont St. Giles, Buckinghamshire, UK). The internal standard for the determination of tigecycline in serum and urine, were received from GE Healthcare (Chalfont St. Giles, Buckinghamshire, UK). Tigecycline reference standard (98.4% chemical purity), 9-aminomycinolic acid-hydrochloride salt reference standard (CL-318614; 97% chemical purity), and N-acetyl-9-aminomycinolic acid (WAY-188749, batch L23566-162) were received from Wyeth Research. All other reagents and chemicals were obtained from commercial sources.

#### Study Design

This open-label, inpatient, multiple-dose tigecycline, single-dose [14C]tigecycline metabolic disposition and mass balance study was performed in healthy men. Eligible subjects were selected on the basis of inclusion/exclusion criteria, medical history, physical examination, and additional procedures outlined in the study protocol. Subjects using any investigational or prescription drug within 30 days of test article administration were excluded from the study. Before initiation of the study, the protocol and consent form were reviewed and approved by an institutional review board. Study subjects were given written informed consent before the screening process was initiated.

#### Dose Administration

The study was initiated with 12 healthy adult male volunteers to ensure six subjects would receive the radiolabeled dose. Each subject received a 100-mg loading dose on the morning of day 1, followed by a 50-mg maintenance dose every 12 h for an additional five doses. On the morning of study day 4, six subjects (ranging from 22 to 40 years of age, with a body weight of 70.1–90.8 kg) each received a single 50-mg dose of [14C]tigecycline, approximately 50 μCi/subject. Each tigecycline dose was administered via a 30-min intravenous infusion.

Throughout the study, subjects received a high fiber meal approximately 2 h before tigecycline administration. Study subjects received two tablets of Fibercan (Wyeth) daily from day 1 through day 13 to facilitate regular bowel movements. The dosing and “inpatient-confinement portion” of the study were conducted at Quintiles (Kansas City, MO). Aliquots of the [14C]tigecycline dosing solution obtained before and after dose administration, dosing vials, and dosing apparatus were shipped to ABC Laboratories for radioactivity analysis.

#### Clinical Sample Collection

Whole blood samples (3 ml) from each subject, for the determination of total radioactivity, were collected into Vacutainer tubes (BD, Franklin Lakes, NJ) containing EDTA within 2 h before [14C]tigecycline dosing and approximately 0.5, 1, 2, 4, 8, 12, 24, 36, 48, 72, 96, 120, 144, 168, 192, 216, and 240 h after the [14C]tigecycline dose. Additional whole blood samples (7 ml) were collected within 2 h before the first tigecycline dose and at the same collection times as the 3-ml whole blood samples. From these whole blood samples, serum was obtained after clot formation and centrifugation of the sample, and used to determine tigecycline and total radioactivity concentrations in serum. Whole blood samples (50 ml) for serum collection were drawn approximately 2 h before the first tigecycline dose, within 2 h before [14C]tigecycline dosing and approximately 1, 4, 8, 24, and 48 h after the [14C]tigecycline dose. These serum samples were stored at −70°C until analyzed for metabolite profiles.

Urine samples were collected within 2 h before the [14C]tigecycline dose and all urine was collected from study subjects after the [14C]tigecycline dose at intervals of 0 to 4, 4 to 8, and 8 to 4 h thereafter up to 240 h. Samples were kept refrigerated during the collection intervals and then were separated into samples for total radioactivity determination, tigecycline concentration determination, and metabolite profiling (samples up to 48 h after the [14C]tigecycline dose), and stored at −8 to −20°C until analysis.

When possible, a fecal sample was collected from subjects before the [14C]tigecycline dose, and all fecal samples up to 240 h after the [14C]tigecycline dose were also collected. Samples were homogenized in 3 volumes of water using a Stomacher (3500; Seward Limited, London, UK) and stored in plastic containers at −5°C until analysis. Aliquots of each sample were analyzed for total radioactivity concentrations and samples collected up to 48 h after the [14C]tigecycline dose were analyzed for metabolite profiles.

For two subjects, there were incomplete fecal collections following the [14C]tigecycline dose, due to noncompliance, and another subject withdrew from the study after sample collections at 24 h. Data from these subjects were not used in the mass balance portion of the study, but were used in the pharmacokinetic analysis and metabolite profiling portions of the study.

#### Radioanalysis

All radioactivity determinations were made using either a Tri-Carb model 3100TR liquid scintillation counter (PerkinElmer, Wellesley, MA) or Beckman/LS6000CS or LS6500 liquid scintillation counters (Beckman Coulter, Inc., Fullerton, CA). For dose, serum, and urine analysis, Ultima Gold scintillation fluid (PerkinElmer) was added to a known volume or weight of sample. Samples were then directly analyzed by liquid scintillation counting. Fecal homogenates were weighed, allowed to dry, and combusted using a model 307 sample oxidizer (PerkinElmer). The resultant [14C]CO2 was trapped in Carborosorb in combination with Permafluor and radioassayed by liquid scintillation counting. For all sample matrices, a quench curve was used to correct counting values to dry dose.

#### Measurement of Tigecycline in Serum and Urine

Serum and urine tigecycline concentrations were determined using validated LC-MS/MS assays. For serum, the assay was validated at concentrations between 10 and 2000 ng/ml. In brief, tigecycline and an internal standard ([r-butyld-9]tigecycline) were extracted from serum using 0.1% trifluoroacetic acid in acetonitrile. The denatured protein was removed by centrifugation and the supernatant was removed and evaporated to dryness at 35°C under a stream of air. Samples were reconstituted in 200 μl and an aliquot was injected for LC-MS/MS analysis using a Thermo Fisher Scientific (Waltham, MA) AQUASIL C18, 50 × 2.1 mm, 5-μm analytical column. Tigecycline and the internal standard were measured using the transition of the positive ions m/z 513.2 and 514.3, respectively. For the quality control samples, the overall precision (CV) was ≤10.5% and accuracy ranged from 104 to 108%.

For urine, the assay was validated at concentrations between 0.20 and 80 μg/ml. In brief, tigecycline and an internal standard in urine were diluted; the samples were mixed and transferred to fresh tubes, and an aliquot was injected for LC-MS/MS analysis using a Thermo Fisher Scientific AQUASIL C18, 100 × 2.0 mm, 5-μm analytical column. For the quality control samples, the overall precision (CV) was ≤11.3% and accuracy ranged from 90.4 to 100%.

#### Metabolite Profiling by HPLC

Individual serum samples collected 1, 4, 8, 12, 24, 36, 48, 72, 96, 120, 144, 168, 192, 216, and 240 h after the [14C]tigecycline dose were analyzed for metabolite profiles.

For two subjects, there were incomplete fecal collections following the [14C]tigecycline dose, due to noncompliance, and another subject withdrew from the study after sample collections at 24 h. Data from these subjects were not used in the mass balance portion of the study, but were used in the pharmacokinetic analysis and metabolite profiling portions of the study.
and 8 h after the [14C]tigecycline dose were analyzed for metabolite profiles. The 24- and 48-h samples were not analyzed because the concentration of radioactivity was too low. Each serum sample was divided into two samples of equal volume (approximately 9 ml each) to provide duplicate analyses. EDTA (final concentration of 40 mM) and 3 volumes of acetone were then added to the samples. Samples were mixed and then centrifuged to remove the denatured protein. The supernatant was transferred to a fresh tube containing 20 μl of glacial acetic acid. The pellet was reextracted with 2 ml of water, 160 μl of 0.2 M EDTA, and 6 ml of acetone and processed as described above. The supernatants were combined and the pellet extraction was repeated. Supernatant samples were evaporated to dryness under a stream of nitrogen at room temperature. The residue was reconstituted in 300 μl of acetonitrile/water (1:9). The reconstituted residue was centrifuged to remove particulates and the supernatant transferred to an autosampler vial. The supernatant was assayed for radioactivity content and analyzed for metabolite profiles by HPLC. Extraction of control serum or fecal samples spiked with varying concentrations of [14C]tigecycline showed that the extraction process resulted in a reproducible percentage of tigecycline in samples being converted to its epimer during the extraction process. Therefore, the amount of epimer in the samples before extraction could be determined based on the amount of epimer in the final serum or fecal extracts. Aliquots (1 ml) of the individual urine samples collected up to 48 h after the [14C]tigecycline dose were transferred to clean tubes and EDTA (final concentration of 40 mM) was added. The pH of the urine remained constant at pH 4.5 to 5.5. Samples were mixed and centrifuged, and the supernatants were transferred to autosampler vials. Separate aliquots of the supernatant were assayed for radioactivity content and analyzed for metabolite profiles by HPLC. Fecal homogenates prepared from samples collected within 48 h after the [14C]tigecycline dose and containing greater than 8000 dpm/g were analyzed for metabolite profiles. Aliquots of the fecal samples (approximately 1 g) were transferred to 15-ml tubes, 3 volumes of acetone were added, and the samples were mixed and centrifuged. The supernatant was transferred to a fresh tube, and the pellet was resuspended with 40 mM EDTA in water and reextracted as described above. The pellet was reextracted in this manner a total of three times and the supernatants were pooled. The combined supernatants were evaporated to dryness under a stream of nitrogen. The residue was resuspended in 500 μl of water and centrifuged to remove particulates, and the supernatant was assayed for radioactivity content and analyzed for metabolite profiles by HPLC. For LC/MS quantitation of metabolites M5 (N-acetyl 9-aminomycinoline) and its epimer M4 in serum and urine samples from three subjects, standard curves were generated for each metabolite in each matrix. In serum, the standard curves ranged from 2.1 to 514.0 ng/ml for M4 and 2.8 to 568.6 ng/ml for M5, whereas in urine, the ranges were 5.7 to 632.5 ng/ml for M4 and 4.3 to 367.8 ng/ml for M5. For quantitation of these metabolites, urine samples were prepared in the same manner as those for metabolite profiling. For the serum samples, EDTA (final concentration of 40 mM) was added to each sample and the samples were vortex-mixed. One volume of acetonitrile was then added to each sample, samples were vortex-mixed, and denatured protein was separated by centrifugation. The supernatant was transferred to a fresh tube and concentrated under a stream of nitrogen to approximately 1 ml. The remaining supernatant was then assayed for M4 and M5 concentrations. HPLC analyses for metabolite profiling were performed using a Waters 2695 Alliance Separation Module (Waters Corp., Milford, MA) with the sample temperature set to 4°C and a Waters model 2487 dual wavelength UV absorbance detector set to monitor 350 nm. The system was in-line with a Gilson 215 liquid handler (Gilson Inc., Middleton, WI) equipped to collect fractions at 20-s intervals. Fractions were collected into 96-well deep-well plates and analyzed for radioactivity using TopCount NXT (PerkinElmer, Shelton, CT). Flo-One analytical software (version 3.65) was used to integrate the radioactive peaks. Separation of tigecycline and drug-derived products was achieved on a Phenomenex Luna C18(2) column (150 × 2.1 mm, 5 μm; Phenomenex, Torrance, CA) equipped with a Phenomenex SecurityGuard guard cartridge (5 μm) using two mobile phases, 10 mM ammonium acetate in water, and acetonitrile. The column was at an ambient temperature of approximately 20°C and the flow rate was 0.2 ml/min. The linear gradient was delivered as follows: 0 to 25 min, from 98% aqueous to 90% aqueous; 25 to 50 min, from 90% aqueous to 70% aqueous; hold at 70% aqueous until 60 min; return to initial conditions at 61 min. Additional HPLC conditions were used to characterize a polar metabolite that was not retained on the C18 column. Separations were accomplished on a Waters Atlantis HILIC Silica column (150 × 2.1 mm, 5 μm) using two mobile phases, 0.02% trifluoroacetic acid in acetonitrile (v/v) and 0.02% trifluoroacetic acid in water (v/v). The column was at an ambient temperature of approximately 20°C and the flow rate was 0.2 ml/min. The linear gradient was delivered as follows: 0 to 10 min, hold at 100% organic; 10 min to 66% organic; 5 min to 34% organic; hold for 15 min; return to initial conditions over 1 min. Metabolite Characterization by Mass and NMR Spectroscopy. The HPLC system and separation conditions used for mass spectrometric analysis were similar to those described for metabolite profiling. MS and MS/MS experiments were performed on a Micromass Quattro Ultima triple quadrupole mass spectrometer (Waters Corp.) equipped with an electrospray source and operated in the positive ionization mode. The electrospray needle potential was set at 2.75 kV and the orifice potential was set at 44 V. The ion source was held at 80°C and the desolvation temperature was 250°C. In MS/MS experiments, the collision-activated dissociation of selected precursor ions was conducted using argon as the collision gas. The collision energy was 30 eV. Serum extracts were analyzed for tigecycline and selected metabolites by LC-MS/MS in the selected reaction monitoring mode to reduce interference from endogenous components. These experiments were conducted with a dwell time setting of 200 ms. The following tigecycline-related components were monitored: tigecycline and its epimer, mz 586→513, hydroxy tigecycline and tigecycline N-oxide, mz 603→585 and mz 603→472; N-desmethyl tigecycline, mz 572→499; tigecycline glucuronide, mz 762→569; 9-aminomycinoline, mz 473→456; and N-acetyl-9-aminomycinoline, mz 515→498. Urine and fecal samples were analyzed for tigecycline metabolites by LC-MS/MS analysis for precursors of product ions characteristic of tigecycline. In addition, potential metabolites of tigecycline were searched for in the LC/MS data for subsequent MS/MS analysis. The site of glucuronidation was further investigated using chemical oxidation of tigecycline and its glucuronide metabolite with Fremy’s salt (potassium nitrosodisulfonate). Fremy’s salt selectively oxidizes phenols, aromatic diols, aminophenols, and diamines to the corresponding quinones (Zimmer et al., 1971). The glucuronide metabolite used for these incubations was isolated from human urine following collection of the HPLC column effluent that contained the glucuronide metabolite. Three sets of incubations with Fremy’s salt were prepared: tigecycline alone, tigecycline plus the isolated glucuronide metabolite, and the isolated glucuronide metabolite alone. Incubations with tigecycline contained 10 μg/ml tigecycline in 10 mM ammonium acetate with 20 μM EDTA, and incubations with the tigecycline glucuronide contained the isolated tigecycline glucuronide with EDTA added (final concentration of 20 μM). Incubations were prepared in duplicate, and to one replicate Fremy’s salt was added (2 ml/mg); to the other replicate nothing was added. Samples were incubated for at least 5 min at an ambient temperature of approximately 20°C before being directly analyzed by LC/MS. Tigecycline glucuronide conjugates were synthesized using modified Koenigs-Knoell reaction conditions (Ag2CO3/MeCN in the presence of EDTA and molecule sieves) with an acyl-protected 1α-bromo-sugar as the donor to generate protected tigecycline glucuronides followed by de-protection in the presence of 0.1 N lithium hydroxide (in methanol/water 1:1). Cochromatography of the synthetic tigecycline glucuronides with tigecycline glucuronide (M7) isolated from human urine was used to determine the desired reaction product. NMR spectroscopy of the synthetic glucuronide was performed with a 500 MHz Bruker DRX spectrometer equipped with a Cap-NMR flow cell probe. Proton NMR spectra were acquired with 128K data points. The heteronuclear single quantum correlation (HSQC) spectrum was acquired with 2K data points in F2 and 256 increments in F1. Heteronuclear multiple bond correlation (HMBC) experiments were conducted with 4K data points in F2 and 512 increments in F1. One of the HMBC spectra was recorded with the J15N setting at 8 Hz and another one was recorded with the J15N setting at 5 Hz for three-bond long-range correlation detection. Pharmacokinetic Analysis. The radioactivity concentration data in serum (expressed in ng-Eq of tigecycline/ml), and the tigecycline serum concentration data for each subject were analyzed by using empirical, model-indepen-
dent pharmacokinetic methods (Jusko, 1992). Peak concentration \((C_{\text{max}})\) was directly determined from the observed data. The apparent terminal-phase disposition rate constant \((\lambda_d)\) for tigecycline in serum was estimated by a log-linear regression of the last three to seven observed concentrations that were determined to be in log-linear elimination by visual inspection. Due to low concentrations and rapid elimination, the concentration of radioactivity in serum could be determined only in samples collected within 24 h of dosing, and therefore, the \(\lambda_d\) for this parameter was estimated using only two or three observed concentrations. The apparent terminal-phase disposition half-life \((t_{1/2})\) was calculated as \(t_{1/2} = 0.693/\lambda_d\).

Since only one dose of \([^{14}\text{C}]\)tigecycline was administered, the serum radioactivity concentration data were analyzed as single-dose data. The single dose area under the concentration-time curve (\(\text{AUC}_{0-\infty}\)) to the last observable concentration \(\left(C_T\right)\) at time \(T\) was calculated using the log-trapezoidal rule for decreasing concentrations and the linear-trapezoidal rule for increasing concentrations. The total single-dose \(\text{AUC}\) was estimated by \(\text{AUC} = \text{AUC}_{0-\infty} + C_T/\lambda_d\). The single dose mean residence time \(\text{(MRT)}\) was calculated as \(\text{AUC} / \text{AUMC} = T_{\text{MRT}}/2\), where AUMC is total area under the first moment curve and \(T_{\text{MRT}}\) is the duration of infusion (30 min).

The unlabeled tigecycline serum concentration data represent multiple-dose administration, and the steady-state \(\text{AUC}\) over one dose interval \(\left(\text{AUC}_{\text{ss}}\right)\), where \(T = 12\ h\) was calculated by using the log-trapezoidal rule for decreasing concentrations and the linear-trapezoidal rule for increasing concentrations. In addition, the steady-state MRT was calculated as \(\text{AUC}_{\text{ss}} \cdot (\text{AUMC}_{\text{ss}} + \tau \cdot \text{AUC}_{\text{ss}})/T_{\text{MRT}}/2\). Systemic clearance \(\text{(CL)}\) was calculated as dose/AUC, and the apparent steady-state volume of distribution \(\left(V_{\text{SS}}\right)\) was estimated as CL \(
\times \text{MRT}.

Calculations. The amount of total radioactivity (dpm) in the urine, feces, and serum was determined by weighting the number of samples by the radioactivity concentration (dpm/g) of each sample. The dose recovered was determined by total dpm in the sample at any given time point, divided by total radioactivity (in dpm) of the dose received by each subject, and multiplied by 100%. These calculations and the calculations of means and standard deviations were performed using Microsoft Excel 2000 software (Microsoft, Redmond, WA). If the dpm in any postdose sample aliquot was less than or equal to 2 times dpm in the predose sample aliquot (background), the value was presented as not detectable (N.D.).

The concentration of tigecycline-related components in serum, urine, and feces was calculated based on the total radioactivity concentrations. These concentrations were converted to tigecycline equivalent using the specific activity of the dose (1.00 \(\mu\text{Ci/mg}\)). Using this value, the concentrations of the specific components were then estimated based on the distribution of radioactivity in the radiochromatograms. These metabolite concentrations reflect only the disposition of the final \([^{14}\text{C}]\)tigecycline dose and do not account for tigecycline or its metabolites remaining from the unlabeled doses. For the concentration determinations of the nonradiolabeled metabolites M4 and M5, the LC/MS peak area from study samples was compared with a standard curve generated using peak areas of control samples spiked with synthetic standards of M4 and M5.

Results

Excretion of Radioactivity. The final \([^{14}\text{C}]\)tigecycline intravenous dose contained 45.9 ± 0.9 \(\mu\text{Ci}\) (mean ± S.D.), equivalent to 45.9 ± 0.9 mg. Recovery of radioactivity in urine and feces was calculated for five subjects (Table 1), with the excluded subject having withdrawn from the study after the 24-h sample collections. For two subjects who had incomplete sample collection, the recovery of radioactivity was <55% and these subjects were not used in the final mass balance calculations. The overall recovery of radioactivity by 240 h after administration of \([^{14}\text{C}]\)tigecycline was 91.8 ± 5.6% (mean ± S.D., \(n = 3\)), with feces being the major route of elimination. The majority of the radioactive dose (87.5 ± 5.1%) was eliminated within 120 h after dosing (Fig. 2).

Pharmacokinetic Analysis. The concentration-time curve and pharmacokinetic parameters for tigecycline in serum following the single tigecycline dose were different from those for radioactivity in serum (Table 2, Fig. 3). The tigecycline concentrations in serum represent tigecycline present from all doses, whereas the radioactivity measurements represent only the final, radiolabeled dose of tigecycline. Tigecycline in serum had a long \(t_{1/2}\) and a large \(V_{\text{SS}}\), and tigecycline concentrations were measurable for 7 to 9 days after the final dose. Tigecycline CL from serum was 22 l/h (0.29 l/h/kg), whereas mean renal clearance \(\left(C_{\text{LR}}\right)\) was 4.6 l/h (data not shown). For

![Fig. 2](https://example.com/fig2.png)
The predominant radiolabeled component was unchanged [14C]tigecycline. Unchanged [14C]tigecycline was the pre-

dominant drug-related component in the serum samples at all time points (Table 3). The values for tigecycline and the epimer of tige-
cycline take into account the amount of tigecycline that was converted to the epimer during the extraction process, as described under Materials and Methods. In serum, M1 (β-butylaminoacetylamino side chain) and the glucuronide conjugates of tigecycline and its epimer (M3 and M2, respectively) increased in abundance, relative to tigecycline, from 1 to 8 h after the radioactive dose. Since the radiolabel had been cleaved from N-acetyl-9-aminominoxylic acid (M5) and its epimer (M4), the metabolites were detected and quantified in serum and urine by LC/MS analysis (Table 4). Therefore, the amount of M4 and M5 detected was the result of all of the tigecycline doses and not just the final radiolabeled dose. Metabolite M5 was detected in all serum samples analyzed at concentrations approximately 5 to 7% of the tigecycline concentrations in the same serum samples. The trace metabolite 9-aminominoxylic acid (M6) was detected by LC/MS analysis of the serum, but was not quantified.

In urine collected up to 48 h after the [14C]tigecycline dose, the predominant radiolabeled component was unchanged [14C]tigecycline (Table 5). The epimer of tigecycline represented 2% of the radioactive dose and increased relative to tigecycline, over time. Some of the epimer may have formed in the bladder, since stability analysis demonstrated that tigecycline degraded to the epimer when incubated in human urine at 37°C (data not shown). In the 0- to 48-h urine samples, M1 represented 6.3% of the radioactive dose, whereas M3 represented 4.1%; M2 was not detected in urine. Within 48 h of the final tigecycline dose, approximately 1.5 mg of unlabeled metabolites M4 and M5 had been excreted in urine (Table 4). Additional trace metabolites in urine, detected by LC/MS, were 9-aminominoxylic acid (M6) and two hydroxy tigecycline metabolites (M7 and M8).

For the fecal samples collected up to 48 h after the [14C]tigecycline dose, tigecycline was the predominant drug-related component (Table 5). The epimer of tigecycline, glucuronide conjugates of tigecycline and its epimer (M4 and M5), and β-butylaminoacetic acid (M1) were all observed in feces. Additional uncharacterized radioactive peaks were observed in some fecal samples and represented less than 2.5% of the radioactive dose. The unlabeled metabolites 9-aminominoxylic acid (M6), N-acetyl-9-aminominoxylic acid (M5), and its epimer (M4) were not detected in any fecal samples.

**Metabolite Structural Characterization.** Figure 5 shows the proposed structures for each of the metabolite peaks observed by HPLC with radioactivity flow detection and/or LC/MS. Tigecycline generated a protonated molecular ion, [M + H]+, at m/z 586. The product ions of m/z 586 mass spectrum included m/z 569, 513, 456, 211, 154, 126, and 86 (Fig. 6; Table 6). Loss of NH3 from [M + H]+ generated m/z 569. The product ion at m/z 513 represented loss of the β-butylamino group from the β-butylaminoacetylamino side chain. Loss of the entire β-butylaminoacetylamino side chain generated m/z 456. The product ions at m/z 211, 154, and 126 originated from the A ring of the tetracycline ring system as indicated in the fragmentation scheme. The product ion at m/z 86 represented the β-butylaminomethylenyl group. This proposed fragmentation scheme was consistent with
the reported fragmentation of structurally similar antibiotics (Kamel et al., 2002).

The epimer of tigecycline generated a $[\text{M} + \text{H}]^+$ at $m/z$ 586. Its product ion of $m/z$ 586 mass spectrum was identical to the mass spectrum for tigecycline (Table 6). Identification as the epimer was made based on its relative retention time being shorter than that of tigecycline.

The $[\text{M} + \text{H}]^+$ for M2 and M3 was observed at $m/z$ 762, which was 176 Da larger than tigecycline (Table 6). Neutral loss of 176 Da generated $m/z$ 586, also the $[\text{M} + \text{H}]^+$ for tigecycline, which indicated a glucuronide of tigecycline. Product ions at $m/z$ 569, 513, 456, 211, 154, and 86 were also observed for tigecycline and did not indicate the site of conjugation. The product ion at $m/z$ 551 was formed by fragmentation of ring B, which indicated that the hydroxy group at either position 10 or 12 was the site of glucuronidation. LC/MS analysis of the reaction product formed after chemical oxidation of tigecycline with Fremy’s salt showed that tigecycline was oxidized to the quinone product for tigecycline, with subsequent oxidation of the hydroxyl group at position 12 (Fig. 7). Consistent with the mechanism of oxidation by the nitrosodisulfonate radical generated from Fremy’s salt, the initial site of reaction was position 7 (Zimmer et al., 1971). Oxidation of the D ring phenol to a quinone required the presence of a free hydroxyl group at position 10 to generate the cyclohexadienone intermedi-

**TABLE 3**

<table>
<thead>
<tr>
<th>Time</th>
<th>t-Butylaminoacetic Acid (M1)</th>
<th>Glucuronide Metabolites (M2 and M3)</th>
<th>Tigecycline Epimer$^a$</th>
<th>Tigecycline$^a$</th>
</tr>
</thead>
<tbody>
<tr>
<td>1 h</td>
<td>9.8 ± 3.2</td>
<td>0.7 ± 1.2</td>
<td>8.3 ± 2.1</td>
<td>81 ± 23</td>
</tr>
<tr>
<td></td>
<td>(14 ± 4.1)</td>
<td>(0.8 ± 1.3)</td>
<td>(12 ± 3.0)</td>
<td>(118 ± 33)</td>
</tr>
<tr>
<td>4 h</td>
<td>17 ± 4.5</td>
<td>12 ± 8.0</td>
<td>6.8 ± 3.7</td>
<td>64 ± 29</td>
</tr>
<tr>
<td></td>
<td>(13 ± 5.5)</td>
<td>(8.4 ± 5.5)</td>
<td>(5.0 ± 2.7)</td>
<td>(47 ± 21)</td>
</tr>
<tr>
<td>8 h</td>
<td>19 ± 6.2</td>
<td>15 ± 10</td>
<td>5.4 ± 2.1</td>
<td>58 ± 28</td>
</tr>
<tr>
<td></td>
<td>(8.2 ± 2.8)</td>
<td>(7.4 ± 5.8)</td>
<td>(2.3 ± 0.9)</td>
<td>(25 ± 12)</td>
</tr>
</tbody>
</table>

$^a$ Corrected for the epimer formed during the extraction process.
The hydroxyl group at position 12 was subsequently oxidized to a keto group, but this would not have occurred without prior oxidation of the D ring phenol to a quinone. After incubations of M3 with Fremy’s salt, no reaction products were observed. If the 10-hydroxyl group were the site of glucuronidation, then no cyclohexadienone intermediate could be formed and no oxidation products of tigecycline glucuronide (M3) would be expected. However, if the 12-hydroxyl group were the site of glucuronidation, then oxidation products of tigecycline glucuronide (M3) would be expected. The inability of tigecycline glucuronide to be oxidized by Fremy’s salt indicated that a free phenolic hydroxyl group was no longer present at position 10 on account of position 10 being the site of glucuronidation. In the ^1H NMR spectrum of the synthetic tigecycline glucuronide, the proton signal for the C10-OH expected at δ 12.3 ppm for tigecycline was absent (data not shown), suggesting a 10-O-glucuronide structure. This suggestion was supported by a three-bond long-range correlation between H1′ (anomeric proton of the sugar moiety at δ 4.47) with C10 (δ 141.7) observed in the HMBC spectrum (Fig. 8). The assignment of C10 at δ 141.7 was confirmed by the three-bond long-range correlation of H8 (δ 8.26) to C10. Therefore, the synthetic tigecycline glucuronide and tigecycline glucuronide isolated from human urine (M3) were both identified as tigecycline 10-O-β-D-glucuronide. Metabolite M2 was proposed to be the epimer of M3 based on its HPLC retention time being earlier than that of M3.

Metabolites M4 and M5 were observed in serum and urine, and only by LC/MS. The [M + H]^+ for these metabolites was observed at m/z 515, which was 71 Da smaller than tigecycline. The product ions of m/z 515 mass spectrum for M5 included m/z 498, 456, 154, 126, and 58 (Table 6). Mass spectral data for M4 and M5 were similar and indicated that the tetracycline ring was intact. The lack of radiochromatographic peaks for M4 and M5 was consistent with loss of the radiolabeled carboxyl group of the t-butylaminocetlylamino side chain. The most likely mechanism for formation of M5 was amide hydrolysis to generate t-butylaminocetlylacidic acid (M1, which contained the radiolabel) and 9-aminoiminocycline (no ^14C label). N-Acetylation of 9-aminoiminocycline was proposed to generate M5. The 58-Da decrease between the m/z 456 product ion and molecular weight (514) was consistent with the presence of a nonradiolabeled acetylaminocycline group and the observation of a m/z 58 product ion for M4 and M5 (Table 6). Confirmation of these proposed structures was obtained by cochromatography of a urine extract spiked with synthetic N-acetyl-9-aminoiminocycline (data not shown). N-Acetylation of 9-aminoiminocycline to form M5 was also demonstrated in human liver cytosol (data not shown). Metabolite M4 was proposed to be the epimer of M5 based on its HPLC retention time being earlier than that of M5. Therefore, M4 and M5 were identified as the epimer of N-acetyl-9-aminoiminocycline and N-acetyl-9-aminoiminocycline, respectively.

9-Aminominocycline (M6) was observed in serum and urine in trace amounts and only by LC/MS due to loss of the radiolabel. The [M + H]^+ for 9-aminoiminocycline was observed at m/z 473. The product ions of m/z 473 mass spectrum included m/z 456 (Table 6), generated from loss of NH3, which was characteristic of
FIG. 5. Tigecycline-related compounds detected in human serum (S), urine (U), and feces (F). * indicates the site of the $^{14}$C label.

Fig. 6. Product ions of $m/z$ 586 mass spectrum for tigecycline.
tigecycline-related compounds. In serum samples, this metabolite was observed by monitoring the $m/z$ 473→456 SRM transition (data not shown). Confirmation of the identification of 9-aminomycinocycline was obtained by comparison with a synthetic standard. 9-Aminominocycline was proposed to have been generated by amide hydrolysis of the $t$-butylaminoacetylamino side chain with radiolabeled $t$-butylaminoacetic acid as a byproduct. Metabolite M1 was observed in serum, urine, and feces, as an early eluting radiochromatographic peak. Isolation of this early eluting M1 peak from human urine and subsequent LC/MS analysis with normal phase HPLC resulted in observation of $[M+H]^+$ and $[MH+CH_3CN]^+$ for M1 at $m/z$ 132 and 173, respectively (data not shown). This indicated a molecular weight of 131, consistent with the identification as $t$-butylaminoacetic acid.

Hydroxylated tigecycline metabolites M7 and M8 were observed in urine in trace amounts. These metabolites produced a $[M+H]^+$ at $m/z$ 602. Mass spectral data for both of the hydroxy metabolites were similar (Table 6). For both M7 and M8, product ions at $m/z$ 529 and 472 were 16 Da larger than the corresponding ions at $m/z$ 513 and 456, respectively, for tigecycline. These data indicated that the tetracycline ring was the site of metabolism rather than the $t$-butylaminoacetylamino group. For M8, the product ions at $m/z$ 211 and 154 indicated an unchanged A ring. Therefore, these metabolites were the product of oxidation of the tetracycline moiety.

**Discussion**

Intravenous administration of tigecycline for 3 days (total dose of 350 mg), followed by a single i.v. dose of $[^{14}C]$tigecycline (50 mg, 50 μCi), was well tolerated. The rationale for preceding the $[^{14}C]$tigecycline dose with multiple doses of nonradiolabeled drug was based upon the ability of tigecycline to bind to bone (Tombs, 1999). This study design was meant to minimize the amount of radioactivity binding to bone while maximizing the recovery of radioactivity.

Another consequence of this study design was that the pharmacokinetic parameters for unlabeled tigecycline in serum were somewhat different from those for radioactivity in serum (Table 2). Tigecycline pharmacokinetics in serum for the current study, including $C_{max}$, AUC, CL, $V_{SS}$, and $t_{1/2}$, were similar to previously reported data under steady-state conditions (Muralidharan et al., 2005). Also consistent with previous data, the concentration versus time curve for tigecycline in serum, as well as for radioactivity in serum, appeared to be biphasic, with rapid distribution into tissues followed by a longer elimination phase, probably caused by the continual release of tigecycline from slowly equilibrating tissues such as bone. However, the

### Table 6

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<thead>
<tr>
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<tr>
<td>M2</td>
<td>27.3</td>
<td>762</td>
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<tr>
<td>M6</td>
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<td>473</td>
<td>456</td>
<td></td>
</tr>
<tr>
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<td>34.7</td>
<td>602</td>
<td>585, 529, 472, 211, 154, 86</td>
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<tr>
<td>M8</td>
<td>38.3</td>
<td>602</td>
<td>585, 529, 472, 86</td>
<td></td>
</tr>
</tbody>
</table>

* Retention times were normalized to retention times of metabolites from serum extracts analyzed by radiochromatography.

![FIG. 7. Product ions of m/z 555 mass spectrum for the tigecycline Fremy’s salt reaction product.](image-url)
After intravenous administration of [14C]tigecycline, unchanged drug was the predominant radiolabeled component in serum, urine, and feces, with more than half of the radioactivity excreted within 48 h of dosing identified as [14C]tigecycline. The epimer of tigecycline was also observed in each matrix, accounting for approximately 5 to 10% of the radioactivity in serum, urine, and feces. The epimer of tigecycline was shown to be a degradant rather than a metabolite, inasmuch as it was formed when tigecycline was incubated in control human serum and urine at 37°C (data not shown). Epimerization at the C-4 position has also been reported for structurally similar antibiotics (Remmers et al., 1963; Nelis and De Leenheer, 1982).

Metabolites M3 (tigecycline glucuronide) and its epimer, M2, were observed in serum, urine (M3 only), and feces. These metabolites represented up to 15% of the radioactivity in serum, and nearly 10% of the dose was excreted as glucuronide conjugates within 48 h of dosing. There are no reports in the literature that identify glucuronidation as a metabolic pathway for antibiotics with similar structures. In fact, these structurally similar antibiotics generally undergo little or no metabolism in humans (Kelly et al., 1961; Eisner and Wulf, 1963; Swarz, 1976; Nelis and De Leenheer, 1981). One exception to this is minocycline, which is metabolized to 9-hydroxyminocycline, N7-demethylminocycline, and N7-demethylminocycline in humans (Nelis and De Leenheer, 1982). The 9-hydroxyminocycline and N-demethylminocycline (site of demethylation not determined) metabolites can each represent up to 8% of the dose in urine (Böcker et al., 1991). Additional metabolites of minocycline have been reported in human urine, but these metabolites have not been characterized (Nelis and De Leenheer, 1982). No N-demethyl metabolites of tigecycline were observed in the current study. Trace amounts of two hydroxy tigecycline metabolites were detected by LC/MS analysis in urine, but the site of hydroxylation and detection by radiochromatography were not possible because the metabolites were present at very low concentrations.

Metabolite M1 (t-butyln-aminoacetic acid) was observed in each biological matrix, accounting for up to 20% of the radioactivity in serum and approximately 8% of the dose was excreted as M1 within 48 h. Metabolism of tigecycline to M1, via amide hydrolysis of the t-butyln-aminoacetylaminogroup, also led to the formation of 9-aminominocycline (M6, not radiolabeled). However, only trace amounts of M6 were detected in serum and urine. The difference in the amount of M1 and M6 observed can be accounted for because a large proportion of the M6 was further metabolized via N-acetylation to N-acetyl-9-aminominocycline (M5) and its epimer (M4). This metabolic pathway was demonstrated using human liver cytosol incubated with M6 (data not shown). Although both M5 and M6 demonstrated some antibacterial activity in vitro, their activities were generally less than that of tigecycline and were effective against a narrower range of infectious organisms (data on file at Wyeth Research). Metabolite M4 was not tested for antibacterial activity since epimerization at the C-4 position in tigecycline and structurally similar antibiotics significantly decreases activity (Zhanel et al., 2004).

It is important to recognize that the excretion data and most of the metabolite profiling performed in this study assess only the disposition of the final [14C]-labeled tigecycline dose, and does not account for tigecycline-related products remaining from the unlabeled doses. Therefore, the estimated concentrations of the tigecycline-related products that were based on radioactivity concentration data may underestimate the actual concentrations of these components in serum, urine, and feces. The exception to this is the concentrations of the unlabeled metabolites M4 and M5, which were determined by LC/MS analysis rather than radioactivity concentrations. The amounts of [14C]tigecycline in serum and urine reported are somewhat lower (generally 10 to 75% less) than the total tigecycline amounts reported. In addition, the contribution of the [14C]tigecycline to the total tige-
cyclophone amount in serum and urine decreased with time; as mentioned above, this is consistent with the last-in, first-out phenomenon.

In summary, following i.v. administration of tigecycline for 4 days (total dose of 400 mg), including a single i.v. dose of [14C]tigecycline (50 mg, 50 μCi), excretion of unchanged tigecycline into feces was the primary route of elimination, and the secondary elimination pathways were renal excretion of unchanged drug, and metabolism to the glucuronide conjugates and the N-acetyl-9-aminominoacycline metabolite. Although unchanged drug was the predominant radiolabeled component in serum, urine, and feces, three radiolabeled metabolites were detected by radiochromatography and a total of eight metabolites characterized by LC/MS.

Acknowledgments. We recognize and thank Julie Chang and Robin Moore for contributions to the metabolite profiling and characterization portions of this study and Oliver McConnell for contributions to the characterization of the glucuronide metabolite using chemical oxidation. We also thank Joan Korth-Bradley for insightful discussions and recommendations. We appreciate the contributions of Ping Cai and Jianxin Gu in identifying the structure of the tigecycline glucuronide using NMR spectroscopy.

References


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