TITLE

Novel Bioactivation Pathway of Benzbromarone

Mediated by Cytochrome P450

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RUNNING TITLE

Novel metabolites of benzbromarone and their cytotoxicity

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Non-standard abbreviations

BBR, benzbromarone; BSTFA, N,O-bis(trimethylsilyl) trifluoroacetamide; CYP, Cytochrome P450; DBBQ, 2,6-dibromobenzoquinone; CAT, 2-ethyl-3-(3-bromo-4,5-dihydroxybenzoyl)benzofuran; DBH, 2,6-dibromohydroquinone; G6P, glucose-6-phosphate; G6PDH, glucose-6-phosphate dehydrogenase; 1'-OH-BBR, 1'-hydroxybenzbromarone; 6-OH-BBR, 6-hydroxybenzbromarone; SIM, selected ion monitoring
ABSTRACT

Benzbromarone (BBR) is a hepatotoxic drug but the detailed mechanism of its toxicity remains unknown. We identified 2,6-dibromohydroquinone (DBH) and mono-debrominated catechol (2-ethyl-3-(3-bromo-4,5-dihydroxybenzoyl)benzofuran; CAT) as novel metabolites of BBR in rat and human liver microsomal systems by comparison with chemically synthesized authentic compounds, and also elucidated that DBH is formed by cytochrome P450 (CYP) 2C9 and that CAT is formed mainly by CYP1A1, 2D6, 2E1 and 3A4. Furthermore, CAT, DBH and the oxidized form of DBH are highly cytotoxic in HepG2, compared with BBR. Taken together, our data demonstrate that DBH, a novel reactive metabolite, may be relevant to BBR-induced hepatotoxicity.
INTRODUCTION

Benzbromarone (BBR), a potent human uric acid transporter I (hURAT1) inhibitor, is effective in the treatment of hyperuricemia and gout, particularly when allopurinol, an international first-line treatment, is ineffective. It has been used as a quite useful drug in Japan for a long time (Masseoud et al., 2005). However, BBR has been associated with hepatotoxicity, including fulminant or fatal liver injury (van der Klauw et al., 1994). Thus, the use of BBR is not approved in the United States and has been withdrawn from the market in Europe (Lee et al., 2008).

Adverse reactions are one of the most important issues in drug development and in the therapeutic usage of drugs during the post-approval stage. Specifically, idiosyncratic adverse drug reactions occur only in a small group of patients who are treated with certain drugs at therapeutic does and are unpredictable (Ulrich, 2007). It is widely accepted that drug-induced idiosyncratic hepatotoxicity is often associated with cytochrome P450 (CYP)-mediated bioactivation (Kalgutkar et al., 2012; Stephens et al., 2014). Thus, BBR could be metabolized to reactive metabolites that cause liver injury.

Early metabolic studies identified that the major metabolites of BBR are 6-hydroxybenzbromarone (6-OH-BBR) formed by CYP2C9 and CYP2C19, and 1'-hydroxybenzbromarone (1'-OH-BBR) produced by CYP3A4 (Kobayashi et al., 2012; McDonald and Rettie, 2007). It has been reported that human hepatocarcinoma HepG2 cells infected with adenovirus vector expressing CYP2C9 significantly increase the cytotoxicity of BBR, suggesting that CYP2C9 is associated with BBR-induced cytotoxicity (Imamura et al., 2011). Thus, the formation of 6-OH-BBR catalyzed by CYP2C9 was thought to play a key role in hepatic toxicity. McDonald et al. proposed that 5,6-dihydroxybenzbromarone, the secondary metabolite from 6-OH-BBR by CYP2C9, and the o-quinone form may be reactive...
metabolites of BBR, although they could not show the formation of such reactive metabolites when BBR was used as a substrate in liver microsomal systems (McDonald and Rettie, 2007). Conversely, Kobayashi et al. reported that 6-OH-BBR was not cytotoxic but that BBR and 1’-OH-BBR exhibited cytotoxicity in human hepatocarcinoma FLC4 cells cultured in micro-space cell in which the expression levels of CYP2C9 and CYP3A4 had been enhanced (Kobayashi et al., 2013). They proposed that 1’-OH-BBR might be the key metabolite that causes liver injury induced by BBR. However, 1’-OH-BBR is formed predominantly by CYP3A4 and it remains unknown why CYP2C9 is primarily responsible for BBR-induced hepatotoxicity.

In previous research, we discovered ipso-substitution as a unique metabolic reaction of a variety of substituted phenols by CYP (Ohe et al., 1994; Ohe et al., 1995; Ohe et al., 1997; Ohe et al., 2000; Tezuka et al., 2007; Nakamura et al., 2011). We hypothesized that BBR, which contains a phenolic hydroxyl group, could be oxidized to 2,6-dibromohydroquinone (DBH) and/or mono-debrominated catechol (2-ethyl-3-(3-bromo-4,5-dihydroxybenzoyl)benzofuran; CAT) through ipso-substitution. The formation of DBH and CAT from BBR might be linked to the mechanism for BBR-induced hepatotoxicity because catechols, hydroquinones, and their oxidized forms are known to be toxic via the formation of covalent bonds with biological macromolecules and the catalytic reduction of oxygen to superoxide and other reactive oxygen species.

The purpose of this study is to elucidate the formation of ipso-substitution metabolites in liver microsomal systems and their cytotoxicity. To this end, we report that these metabolites may be relevant to the hepatotoxicity induced by BBR.
MATERIALS AND METHODS

**Materials.** NADP$^+$ was obtained from Wako Pure Chemical Industries, Ltd. (Osaka, Japan). Glucose-6-phosphate (G6P) and G6P dehydrogenase (G6PDH) were obtained from Roche Diagnostics (Basel, Switzerland). Hydroquinone and $N,O$-bis(trimethylsilyl)trifluoroacetamide (BSTFA) were obtained from Tokyo Chemical Industry Co., Ltd. (Tokyo, Japan). All other reagents used for metabolism and toxicity experiments were of analytical grade.

**Synthesis of expected metabolites.** We chemically synthesized *ipsosubstituted metabolites, CAT, DBH and the oxidized form of DBH (DBBQ), and the known main metabolite, 6-OH-BBR, as authentic standards for metabolite identification and/or cytotoxicity studies. Briefly, for the synthesis of CAT, vanillin was brominated at the 3-position with Br$_2$ followed by methylation and oxidation to form 3-bromo-4,5-dimethoxybenzoic acid. After treatment with SOCl$_2$, Friedel-Crafts acylation with 2-ethylbenzofuran was directed to the 3-position of the benzofuran ring. Then, the demethylation reaction of the two methoxy groups with BBr$_3$ produced CAT. DBH was obtained via regioselective bromination of 4-methoxyphenol with benzyltrimethylammonium tribromide (Kajigaeshi et al., 1987). DBBQ was directly prepared by the oxidation of DBH using Oxone$^\circledast$ (potassium peroxymonosulfate). 6-OH-BBR was synthesized according to the procedure reported by McDonald et al. with minor modifications (McDonald and Rettie, 2007). More detailed procedures and analytical data for synthesized compounds are shown in Supplemental Methods.

**Liver microsomal incubation.** Phenobarbital-treated SD rat and pooled human (200 donors) liver microsomes were purchased from XenoTech, LLC (Kansas, US). The incubation mixture containing rat or human liver microsomes (1 mg protein/mL), BBR (100 $\mu$M), MgCl$_2$ (3 mM), G6P (10 mM), and G6PDH (1 U/mL) with or without tienilic acid (10 $\mu$M)
in 1.5 mL of 0.1 M K-Pi buffer (pH 7.4) was preincubated for 3 min at 37˚C. The reaction was initiated by adding NADP⁺ (1 mM) and incubated for 30 min (rat) or 120 min (human) at 37˚C. One milliliter of the incubation mixture was treated with 2 mL of ice-cold ethyl acetate to stop the reaction and to extract the products, and the organic phase was separated. The organic phase was dried over anhydrous Na₂SO₄ and concentrated by N₂ flushing. The products formed were trimethylsilylated with BSTFA and pyridine. After the removal of the excess BSTFA by N₂ flushing, the residue was dissolved in a small amount of acetone and analyzed by GC-MS (SHIMADZU GC-MS-QP5050A, capillary column HP-1 0.32 mm × 30 m; J & W Scientific). The injection temperature was 260˚C. The initial column temperature was 90˚C for 3 min; then the temperature was raised at intervals of 10˚C/min to 250˚C, followed by an isothermal hold at this temperature. On the other hand, 0.5 mL of the incubation mixture was treated with 1 mL of acetonitrile : methanol (2:1) to stop the reaction. All samples were centrifuged at 10,000 × g for 10 min, and the supernatant was analyzed by LC-MS (Agilent Technologies 6120) Chromatographic separations were performed on XBride™Shield RP18 column (Waters, 3.5 μm 2.1 × 150 mm) at a flow rate 0.2 mL/min under liner isocratic conditions of acetonitrile/water/HCOOH (60/40/0.1). The eluent was introduced directly into the mass spectrometer via electrospray ionization in the negative ion mode.

**Incubation with CYP isozymes.** Recombinant human CYP isozymes (EasyCYP Bactosomes) were purchased from Cypex, Ltd. (Dundee, UK). Incubation was performed under the same conditions used for liver microsomes described above, except that the incubation mixture contained 0.1 μM CYP isozymes, 0.1 mM BBR, 4 mM MgCl₂, 4 mM G6P, 1 U/mL G6PDH, and 0.4 mM NADP⁺.

**Cell culture.** Human hepatocarcinoma HepG2 cells were obtained from ATCC (Virginia,
US). The HepG2 cells were maintained in Minimum Essential Media (GIBCO) containing 5% fetal bovine serum (GIBCO) and 1% penicillin-streptomycin (SIGMA) in a 5% CO₂ incubator with a humidified atmosphere at 37°C.

**Cytotoxicity assay.** Cell toxicity was evaluated by a trypan blue dye exclusion test. HepG2 cells were seeded (1×10⁶ cells/mL/well) on a 96-well plate (IWAKI). After 24 hr of incubation, the cells were treated with BBR, 6-OH-BBR, CAT, DBH, and DBBQ at 0.4 μM. After being incubated with the drugs for 24 hr, the viable cells were counted according to the dye exclusion method using Vi-Cell™ (Beckman Coulter).
RESULTS

Metabolism of BBR by rat or human liver microsomes. BBR was incubated with phenobarbital-treated rat liver microsomes or pooled human liver microsomes. The metabolite extracts were analyzed by LC-MS for 6-OH-BBR and CAT and analyzed by GC-MS after trimethylsilylation for DBH. All of the metabolites were identified on the basis of retention time and the m/z peak ratio in selected ion monitoring (SIM) mode, compared with the synthesized chemical standards (6-OH-BBR, CAT, and DBH). The quantification results including control incubations (NADP⁺ (-) and microsomes (-) ) are shown in Table 1. The results demonstrate that the novel metabolites CAT and DBH were formed with rat and human hepatic microsomes. Not all of the metabolites were detected when NADP⁺ or microsomes were omitted from the complete system.

Metabolism of BBR by isozymes of human CYP. BBR was incubated with various human CYP isozymes expressed in Escherichia coli (bactosomes). The metabolite extracts were analyzed in a similar manner to the metabolic study with liver microsomes, and the amounts of formed metabolites were determined (Table 2). It was found that the formation of 6-OH-BBR is catalyzed by CYP1A1, CYP2C8, CYP2C9, CYP2C19 and CYP2D6. CYP1A1, CYP2D6, CYP2E1 and CYP3A4 were found to catalyze the formation of CAT, while DBH was formed by only CYP2C9.

Effect of CYP2C9 inhibitor on the BBR metabolite formation. Additionally, we investigated the effect of tienilic acid, a mechanism-based inhibitor specific for CYP2C9, on the formation of three metabolites from BBR in human liver microsomes to confirm the contribution of CYP2C9. BBR was incubated with human liver microsomes in the presence of 10 μM tienilic acid, and the amounts of formed metabolites were determined. As shown in Supplemental Figure S1, the formation of 6-OH-BBR was markedly inhibited by tienilic acid,
and the formation of DBH was also significantly inhibited. In contrast, tienilic acid had no inhibitory effect on the formation of CAT from BBR.

Cytotoxicity of BBR and BBR metabolites in HepG2 cells. The cell toxicities of BBR, 6-OH-BBR, CAT and DBH in the hepatocyte cell line (HepG2 cells) were assessed by the use of the trypan blue dye exclusion method after 24 hr of drug exposure. In addition, based on the possible significance of DBH for the bioactivation process of BBR, the toxicity of the oxidized form of DBH, DBBQ, was also evaluated because DBH is thought to be easily oxidized to DBBQ during incubation, although DBBQ can never be detected in the extracts from the microsomal system because of its susceptibility to bind to macromolecules. The cell toxicities of the compounds at various concentrations are demonstrated in Supplemental Figure S2. As shown in Fig. 2, the ipso-substituted metabolites CAT, DBH and DBBQ induced significant cytotoxicity at the lowest concentration (0.4 μM) and, they exhibited more potent toxicity than did 6-OH-BBR. On the other hand, BBR was not cytotoxic at this concentration.
DISCUSSION

The exact mechanisms leading to the hepatocellular toxicity of BBR, a therapeutically useful uricosuric drug, are not yet understood. The present study is an attempt to elucidate the cause of BBR-induced liver toxicity by proposing a novel bioactivation mechanism for BBR. First, we hypothesized that BBR, which contains a phenolic hydroxyl group, could be subject to an *ipso*-substitution metabolism, which we had previously discovered as a unique metabolic pathway catalyzed by CYP, and oxidized to DBH and/or CAT.

As shown in Table 1, DBH and CAT were formed in rat and human microsomal systems in NADPH- and microsome-dependent manners, as determined from comparison with the synthetic standards, although the formation amounts were much lower than that of 6-OH-BBR. The production of CAT and DBH may be underestimated because the recovery of these metabolites from the microsomal incubation mixtures was shown to be low (data not shown) probably due to the covalent binding of the oxidized forms (quinone forms) with macromolecules such as proteins.

Isozyme profiling with recombinant CYP bactosomes clarified that the formation of 6-OH-BBR is catalyzed by CYP1A1, CYP2C8, CYP2C9, CYP2C19 and CYP2D6. This result is consistent with the previous reports that CYP2C9 and CYP2C19 are the major isozymes capable of catalyzing the formation of 6-OH-BBR (Kobayashi et al., 2012; McDonald and Rettie, 2007). CYP2C9 was also responsible for the formation of DBH from BBR. The additional analysis using the CYP2C9 specific inhibitor further confirmed that CYP2C9 contributes to 6-hydroxylation of BBR and *ipso*-substitution metabolism to generate DBH. There have been some reports that CYP2C9 is primarily responsible for BBR-induced liver toxicity. Iwamura et al. reported that HepG2 cells infected with adenovirus vector expressing CYP2C9 significantly increase the cytotoxicity of BBR (Imamura et al., 2011).
al. suggested that a catechol metabolite, 5,6-dihydroxybenz bromarone, which is formed via 6-OH-BBR, and its o-quinone form may be reactive metabolites and that the sequential metabolism is catalyzed by CYP2C9 (McDonald and Rettie, 2007), although they did not show the formation of such reactive metabolites directly from BBR in liver microsomal systems, and their cytotoxicity has not been elucidated.

The current study demonstrated that 6-OH-BBR was cytotoxic to HepG2 cells but the toxicity was not remarkable. Conversely, novel metabolites, DBH and CAT showed more potent cytotoxicity than BBR and 6-OH-BBR, suggesting that the ipso-substituted metabolites may possibly be reactive metabolites. In addition, the cytotoxicity of the quinone form of DBH, DBBQ, was also examined because it is thought that DBH is easily oxidized to DBBQ in liver microsomal systems. As a result, DBBQ also exhibited potent cytotoxicity that is comparable to that of DBH and CAT. Taking into consideration that DBH is mainly formed by CYP2C9, the contribution of DBH and DBBQ to BBR-induced hepatotoxicity, in particular, might be important.

As described above, Kobayashi et al. proposed that the metabolic activation via 1’-OH-BBR formed by CYP3A4 might be responsible for BBR-induced hepatotoxicity (Kobayashi et al., 2013), but the mechanism is not yet clear and it remains unknown why CYP2C9 enhances the cytotoxicity of BBR. The current study clearly presents an alternative mechanism for BBR-induced hepatotoxicity activated by CYP2C9.

In conclusion, we have demonstrated the formation of reactive metabolites in liver microsomal systems via an ipso-substitution metabolism of BBR, which is a novel metabolic pathway for BBR. We consider that the formation of DBH or DBBQ might be one of the reasons for hepatocellular toxicity induced by BBR. Additional studies to investigate the detailed mechanism for BBR-induced hepatotoxicity are in progress.
AUTHORSHIP CONTRIBUTIONS

Participated in research design: Ohe, Kitagawara, Takahashi, Nakamura, and Mashino.

Conducted experiments: Kitagawara and Tachibana.

Contributed new reagents or analytic tools: Kitagawara, Ohe, and Tachibana.

Performed data analysis: Kitagawara and Ohe.

Wrote or contributed to the writing of the manuscript: Ohe, Kitagawara, and Mashino.
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Ohe T, Mashino T, and Hirobe M (1997) Substituent elimination from *p*-substituted phenols by cytochrome P450: *ipso*-substitution by the oxygen atom of the active species. *Drug Metab Dispos.* **25:** 116-122.


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FIGURE LEGENDS

Fig. 1. Proposed metabolic pathway of benzbromarone.

Fig. 2. Cytotoxicity of BBR and BBR metabolites at 0.4 μM in HepG2 cells. Cell viability was assessed by the use of the trypan blue dye exclusion method after 24 hr of drug exposure. The data are expressed relative to control incubations containing 0.1% DMSO alone. Each value represents the mean ± S.D. of six determinations from two separate experiments. ***P < 0.001 and *P < 0.05 compared with the control incubations. †††P < 0.001 and ††P < 0.01 versus the incubations with 6-OH-BBR.
**TABLE 1**

Formation of metabolites from BBR by rat or human liver microsomal system (nmol/nmol P450)

<table>
<thead>
<tr>
<th>System</th>
<th>6-OH-BBR</th>
<th>CAT</th>
<th>DBH</th>
</tr>
</thead>
<tbody>
<tr>
<td>Rat liver microsomes</td>
<td>3.05 ± 0.58</td>
<td>1.26 ± 0.15</td>
<td>0.07 ± 0.01</td>
</tr>
<tr>
<td>Human liver microsomes</td>
<td>82.0 ± 8.14</td>
<td>0.40 ± 0.18</td>
<td>0.26 ± 0.05</td>
</tr>
<tr>
<td>NADP⁺ (-)</td>
<td>N.D.</td>
<td>N.D.</td>
<td>N.D.</td>
</tr>
<tr>
<td>Microsomes (-)</td>
<td>N.D.</td>
<td>N.D.</td>
<td>N.D.</td>
</tr>
</tbody>
</table>

Each value represents the mean of three samples ± S.D.

N.D.: not detected.
### TABLE 2

Formation of metabolites from BBR by recombinant human CYP isozymes (nmol/nmol P450)

<table>
<thead>
<tr>
<th>System</th>
<th>6-OH-BBR</th>
<th>CAT</th>
<th>DBH</th>
</tr>
</thead>
<tbody>
<tr>
<td>CYP1A1</td>
<td>23.9 ± 2.08</td>
<td>0.75 ± 0.02</td>
<td>N.D.</td>
</tr>
<tr>
<td>CYP1A2</td>
<td>N.D.</td>
<td>N.D.</td>
<td>N.D.</td>
</tr>
<tr>
<td>CYP2B6</td>
<td>N.D.</td>
<td>N.D.</td>
<td>N.D.</td>
</tr>
<tr>
<td>CYP2C8</td>
<td>60.4 ± 3.3</td>
<td>N.D.</td>
<td>N.D.</td>
</tr>
<tr>
<td>CYP2C9</td>
<td>398 ± 10</td>
<td>N.D.</td>
<td>0.15 ± 0.02</td>
</tr>
<tr>
<td>CYP2C19</td>
<td>193 ± 33</td>
<td>N.D.</td>
<td>N.D.</td>
</tr>
<tr>
<td>CYP2D6</td>
<td>15.3 ± 1.2</td>
<td>1.13 ± 0.07</td>
<td>N.D.</td>
</tr>
<tr>
<td>CYP2E1</td>
<td>N.D.</td>
<td>0.66 ± 0.07</td>
<td>N.D.</td>
</tr>
<tr>
<td>CYP3A4</td>
<td>N.D.</td>
<td>0.88 ± 0.00</td>
<td>N.D.</td>
</tr>
</tbody>
</table>

Each value represents the mean of three samples ± S.D.

N.D.: not detected.
Fig. 1

Benzbromarone (BBR) → 6-Hydroxylated to 6-Hydroxybenzlbromarone (6-OH-BBR)

via Ipso-substitution

2,6-Dibromohydroquinone (DBH) → 2,6-Dibromobenzoxquinone (DBBQ)

via 2e⁻ reduction

2-Ethyl-3-(3-bromo-4,5-dihydroxybenzoyl)benzofuran (CAT)
Fig. 2